# Rare Earth and Other Critical Element Concentrations in the Sentinel Butte Formation, Tracy Mountain, North Dakota

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#### Abstract

High rare earth element concentrations (>300 ppm) were found in a thin lignite bed at Tracy Mountain during reconnaissance sampling in southern Billings County, North Dakota in 2016. Tracy Mountain is a series of buttes comprised of alternating beds of sandstone, siltstone, mudstone, claystone, lignite, and clinker representing roughly the lower two-thirds of the Sentinel Butte Formation of the Fort Union Group (Paleocene). A dozen carbonaceous beds and bentonites were traced across the study area. In total, 15 geologic sections were measured across Tracy Mountain and the surrounding area, and 169 rock samples were submitted for critical element ICP-MS analysis, most of which (155) were lignites and carbonaceous claystones or mudstones. Tracy Mountain samples were analyzed for the rare earth elements including yttrium and scandium, 26 additional elements deemed critical to the economic security of the United States, and noncritical elements molybdenum and thorium. Between 21 and 169 analyses of each element were acquired, with the more economically promising elements receiving more focus. Extensive sampling revealed that high rare earth concentrations (>300 ppm) occur in multiple thin carbonaceous beds over an area of approximately 25 acres, with multiple beds exhibiting spot concentrations over 1000 ppm on a whole coal basis. The highest rare earth concentrations occur in the stratigraphically highest carbonaceous beds, carbonaceous beds immediately below bentonites, and in carbonaceous beds associated with white, greasy claystones and mudstones. In samples with rare earth element concentrations >300 ppm, the high-demand elements dysprosium, neodymium, scandium, and terbium make up 25.7% of the total rare earth elements. Magnesium, zirconium, molybdenum, vanadium, gallium, and uranium are also enriched at Tracy Mountain relative to other US coals. Many critical elements, most notably gallium, vanadium, and uranium, are strongly correlated with the rare earths and may have become enriched via the same pathways. Establishing the presence of these elements in North Dakota lignites and the tendency for each to co-occur has implications in identifying the feedstocks, target mineral commodities, and extraction methods for any future commercial production.

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#### Introduction

The North Dakota Geological Survey began a critical elements investigation in the fall of 2015. Rock samples (primarily lignites) were collected for rare earth element (REE) analysis from outcrops in the Little Missouri River badlands in southwestern North Dakota. During the first two years of the study, 65 geologic sections were measured, and 352 samples were analyzed for rare earth element concentrations. The results were reported in Kruger and others (2017), which documented the first detailed occurrences of rare earth enrichment in North Dakota. Murphy and others (2018) reported on high concentrations and lateral variability of REE within lignites at Logging Camp Ranch in Slope County after measuring 20 additional geologic sections and submitting 113 samples for analysis. Since 2018, 1,056 samples have been analyzed and 184 additional geologic sections have been measured. Additionally, in recent years the rock samples have been routinely analyzed for other select critical elements. The primary study area encompasses a 30 x 150 mile (9 x 46 km) rectangle stretching from the North Dakota-South Dakota border in Bowman County to Tobacco Garden Creek in McKenzie County. More recently, the study has been expanded to include central and south-central North Dakota (Figure 1).

Samples were collected across an 1,800-foot (550 m)-long stratigraphic interval from the Fox Hills Formation (Cretaceous) through the Arikaree Formation (Miocene), encompassing 10 of the 12 bedrock formations that are exposed at the surface in western and south-central North Dakota. The focus of the study has been the coal-bearing rocks of the Fort Union Group (Figure 2). The vast majority of analyzed



▲ Figure 1. North Dakota Geological Survey critical element study sample sites. Existing mines are included as points of reference.

samples have been organic-rich rocks (91%), either lignites (72%) or carbonaceous claystones and mudstones (19%). Volcanic ash (tuffs) or altered volcanic ash (tonsteins and bentonites), clinker (scoria), natural coal ash, both iron and manganese nodules, and sandstone concretions have also been analyzed (Table 1). The majority of samples have come from five areas: west-central Billings County, Tracy Mountain, Logging Camp Ranch, the type section of the Slope Formation, and Mud Buttes.



Number of Samples Analyzed by Stratigraphic Unit

◄ Figure 2. The number of samples analyzed for critical elements for the entire NDGS study by stratigraphic position. The Fort Union Group consists of the rocks from the Ludlow Formation up through the Sentinel Butte Formation.

**Table 1.** The number of samples analyzed for REE and other critical elements by lithology.

Entire Project (2015-2021)	Tracy Mtn. (This Report)	Lithology
1,089	110	Lignites
292	45	Carbonaceous clay/mudstones
25	1	Claystones and mudstones
25	7	Bentonites
22		Tonsteins
22	4	Nodules or concretions
20		Volcanic tuffs
12	2	Natural coal ash
11		Sandstones and siltstones
3		K/Pg ejecta
1,521	169	Total

#### Fieldwork

Tracy Mountain consists of roughly a half dozen flat-topped buttes or mesas rising 200 to 300 feet (60-90 m) above the surrounding badlands topography in south-central Billings County, approximately 13 miles (21 km) southwest of the town of Belfield (Figure 3). This butte complex trends from the southwest

to the northeast covering approximately 1,160 acres (4.7 km<sup>2</sup>) in portions of sections 9, 10, 15, and 16 (Township 138N, Range 101W) and has a maximum elevation of approximately 2,900 feet (884 m).

Ned Kruger first measured a geologic section and collected lignite samples for rare earth analysis along the southeast corner of the Tracy Mountain complex in the fall of 2016 (measured section 7 of this report, section 37 of Kruger et al., 2017). The site was chosen for study due to the presence of an old upland surface and the excellent outcrops exposed along the butte slopes. Samples of the upper coal had high rare earth element concentrations in excess of 300 parts per million (ppm) on a whole rock basis (the economic threshold suggested by the U.S. Department of Energy). As a result, an additional section was measured (section 4 of this report, section 38 of Kruger et al., 2017) and two more lignite samples were collected and submitted for analysis in 2017, those samples also contained high concentrations. In 2018, a third geologic section (section 2) was measured 1,700 feet (520 m) to the northwest of measured section 7 and a dozen more rock samples were collected in the upper coals. Consistently high critical element



▲ Figure 3. The topography of the Tracy Mountain area and the locations of the 15 measured sections and sample sites in this study. The exact locations of measured sections 10 and 11 were not plotted on this map.

concentrations in those well-exposed, laterally continuous, upper carbonaceous beds led to the expansion of the project at Tracy Mountain. Previous investigations into the lateral extent of REE enrichment zones had been limited by the exposure of outcrops or the extents of the lignites themselves (Murphy et al., 2018). In 2020 and 2021, 13 more geologic sections were measured and 150 additional rock samples were collected. The 15 geologic sections plot along a seven-mile-long (11 km), two-to-three-mile-wide (3-5 km) transect centered on Tracy Mountain (Figure 3). Measured sections 1-9 were concentrated around the perimeter of the largest mesa in the Tracy Mountain complex (the northwest quarter of section 10, T138N, R101W). Most of the measured sections on this mesa are spaced 300 to 500 feet (90-150 m) apart (Figure 3). Two additional geologic sections are located 0.6 miles (1 km) southwest of the main butte and four sections are located on isolated buttes or uplands beyond the Tracy Mountain complex; one approximately two miles (3 km) to the northwest, another four miles (6 km) to the southwest, and two more roughly three miles (5 km) to the northeast (Figure 3).

A team of three geologists typically performed the fieldwork with one measuring section and the other two collecting the rock samples. Whenever possible, geologic sections were measured along steep, poorly vegetated slopes to minimize the amount of colluvium obscuring the underlying lithologies. A pick was used to dig down to fresh rock so that bed thickness and characteristics of the freshly exposed rock could be recorded in a field notebook. All rock samples were collected in approximately three-inch-thick (8 cm) stratigraphic intervals after a minimum of six inches (15 cm) was removed from the outcrop in order to minimize the collection of weathered rock. A typical sample weighed approximately 3.3 pounds (1,500 g) and was stored in a gallon-size Ziplock bag. Samples were generally taken from the top three inches (8 cm) of a bed or wherever the most carbonaceous rocks were present.

In total, 175 rock samples were collected from the Tracy Mountain study site. The vast majority of these were either from lignites (116 samples) or carbonaceous claystones or mudstones (45 samples). Additionally, one sample was collected from a mudstone, seven samples were collected from bentonites, four from iron-oxide nodules, barite nodules, or sandstone concretions, and two samples were taken of coal ash beneath clinker deposits. Carbonaceous beds in close proximity to the base of bentonites were specifically targeted for sampling.

#### **Laboratory Analysis**

Upon return from the field, rock samples were split into a 2.2 pound (1,000 g) sample that was shipped to Standard Laboratories in Casper, Wyoming and a 1.1 pound (500 g) sample that was archived in the Geological Survey warehouse. Tracy Mountain rock samples were submitted to the laboratory for analysis in 2016, 2017, 2018, 2020, and 2021. In total, 169 of the 175 samples collected were submitted to the laboratory for analysis (Table 1).

The initial focus of the Geological Survey's critical elements project was on rare earths. In 2016, all rock samples, including the first nine samples collected from Tracy Mountain, were analyzed only for rare earth element concentrations (Kruger et al., 2017). The analyses included the fourteen stable lanthanide series elements (lanthanum, cerium, praseodymium, neodymium, samarium, europium, gadolinium, terbium, dysprosium, holmium, erbium, thulium, ytterbium, and lutetium) and yttrium. All concentrations were reported on a whole rock basis. The NDGS added scandium to its analyses in 2017, and it is considered a REE in this report.

On May 18, 2018, the Secretary of the Interior finalized a list of 35 mineral commodities deemed critical: aluminum (bauxite), antimony, arsenic, barite, beryllium, bismuth, cesium, chromium, cobalt, fluorspar, gallium, germanium, graphite (natural), hafnium, helium, indium, lithium, magnesium, manganese, niobium, the platinum group metals, potash, the rare earth elements group, rhenium, rubidium, scandium, strontium, tantalum, tellurium, tin, titanium, tungsten, uranium, vanadium, and zirconium (Interior Dept., 2018). Subsequently, the NDGS expanded the scope of its investigation.

From 2018 on, in addition to REE, Tracy Mountain samples were also analyzed for some of the additional critical elements. The NDGS was also able to obtain additional critical element analyses on some of the previous year's submissions for which the lab had retained sufficient amounts of sample material. Initially, those samples were analyzed for a subset of 26 critical elements; antimony (analyzed for in 81 samples), arsenic (73 samples), barium (81), beryllium (34), bismuth (21), cesium (34), chromium (108), cobalt (55), gallium (128), germanium (130), hafnium (137), indium (21), lithium (34), magnesium (127), manganese (24), niobium (126), rubidium (24), strontium (34), tantalum (108), tellurium (21), tin (21), titanium (34), tungsten (154), uranium (157), vanadium (126), and zirconium (127) along with the noncritical elements molybdenum (157) and thorium (24). Molybdenum, while not considered a critical element in the United States, is a potential value-added product from coal and has been the subject of past commercial exploration in North Dakota lignites, along with uranium and germanium (Murphy, 2005). Thorium, by contrast, is a troublesome contaminant in traditional rare earth ores and can be difficult to separate from the REEs during refinement. Its presence is a consideration in the economic assessment of any REE ore. Due to limitations of the analytical budget, all samples were not analyzed for all 28 elements plus the rare earths. Criteria for which elements to investigate were based upon concentrations relative to lignites from elsewhere in North Dakota, the U.S., and an economic assessment of each element by Dai and Finkelman (2018). In general, more promising elements received more analyses, although some elements considered less promising when produced from coal received additional focus when concentrations were among the highest reported from a North Dakota lignite.

Laboratory analyses for all REE, excluding scandium from six pre-2017 samples, were obtained from 72 of the 169 samples, including all for which  $\sum REE > 300$  ppm. For the other samples, it was determined that reliable estimates of total REE concentrations could be made by analyzing just seven (lanthanum, cerium, neodymium, gadolinium, erbium, yttrium, and scandium) of the sixteen REEs. These estimates appear in italics in Appendix A. Estimates were calculated using linear regression formulas based on the trendlines from scatter plots of each unknown element compared to its closest-affiliated element(s) with a known concentration, obtained from a database containing a full suite of REE analyses from 828 samples from this and previous studies. The linear regressions were found to produce total REE concentrations within a deviation range of -2.6 to 1.8% from the laboratory measurements and an average deviation of  $\pm 0.37\%$ . (Kruger, 2020).

Throughout the project the focus has remained on critical elements. Major ions, including sodium, calcium, potassium, and sulfate were not analyzed to enable funding to be focused on determining the extent of high critical element concentrations.

#### Geology

Tracy Mountain and the surrounding area consist of alternating beds of claystone, mudstone, siltstone, sandstone, lignite, and clinker (scoria) of the Sentinel Butte Formation, Fort Union Group (Paleocene) (Figures 4 and 5, Appendix A). A six-foot (1.82 m) thick exposure of the HT Butte lignite, the base of which marks the contact of the Sentinel Butte Formation with the underlying Bullion Creek Formation, is present in a ravine in the northeast quarter of section 10. For several miles to the north of Tracy Mountain, the HT Butte lignite has burned, forming low, clinker-capped knolls and ridges at an approximate elevation of 2,600 feet (792 m) above sea level. The HT Butte bed is also present at an elevation of approximately 2,650 feet (808 m) at the base of a ravine below measured section no. 13 (Figure 5). The HT clinker caps ridges and small buttes for several miles (kilometers) to the south of this location. The HT Butte lignite is approximately 14 feet (4.3 m) thick in this area and is traceable through the shallow subsurface on gamma ray logs from coal exploration holes and oil wells. The roughly 300 feet (90 m) of Sentinel Butte strata that comprise Tracy Mountain represents approximately 60% of the total formation thickness in this area. The Sentinel Butte Formation is 512 feet (156 m) thick at Square Butte in Golden Valley County (Murphy et al., 1993, figs. 51 and 97).

The HT Butte bed is the thickest lignite in the Tracy Mountain complex with an exposed thickness of six feet (1.8m). None of the other coals in this complex are more than four feet (1.2 m) thick, although buttes to the north (measured section nos. 14 and 15) contain a 6.5-foot (2.0 m) lignite. More than a dozen thin lignites and carbonaceous claystones are exposed on the slopes of Tracy Mountain or in adjacent drainages. For the purposes of this study, these thin lignite and carbonaceous beds were given letter designations A thru S in descending order from the top of the main mesa in the study area (Figure 4). The HT sample designation was used for samples collected from the HT Butte bed. Additionally, a generalized stratigraphic column was constructed for Tracy Mountain and the positions of the upper carbonaceous beds (A-K) were plotted on it (Figure 6). Lithologic description, bed thickness, and sample position are also noted for these carbonaceous beds. Thin lignites are typically difficult to trace laterally for any distance because they can suddenly pinch out, but many of these thin coals could be traced throughout the relatively closely spaced measured sections (nos. 1-9). While traceable, many of these thin coals transitioned from lignite into carbonaceous claystones and mudstones. Thin lignite layers and/or lignite stringers were typically present in these carbonaceous beds (Figures 7-9).

More than a dozen gray to bluish-gray swelling claystones are present in the Sentinel Butte Formation at Tracy Mountain (Figures 4, 5 and 10). These claystones have an irregular, granular weathering surface that results from the swelling and shrinking that occurs when the clays alternately hydrate and dehydrate. In the field, this weathering surface is typically referred to as popcorn texture (Figure 11). The gray to bluish-gray claystones with the well-developed popcorn surface were assumed to be bentonites, but were not confirmed as such by petrographic analysis (Forsman, 1985). Several of these "bentonites" were traceable through measured section nos. 1-9 (Figure 4). Lignites and carbonaceous mudstones that were overlain by bentonites were specifically targeted for sampling (Figures 12 and 13). Forsman (1985) determined that Sentinel Butte claystones and mudstones are typically composed of a mixture of montmorillonite, illite, chlorite, and kaolinite. As a result, Sentinel Butte claystones and mudstones typically contain a swelling clay component.



▲ Figure 4. A geologic cross-section of measured sections 1-9 in the Tracy Mountain study. See Figure 3 for the location of the measured sections.



▲ Figure 5. A geologic cross-section of measured sections 1-15 in the Tracy Mountain study. See Figure 3 for the location of the measured sections.

Tracy Mountain Generalized Stratigraphic Column

Main butte (Measured sections 1 through 9) Upper 100 ft / 30 meters

Eleva	tion					
ft.	m		Bed	Bed th	ickness	Lithologic Description
2900	ſ			in	cm	
			Α	0 to 2	0 to 5	Carbonaceous mudstone, dark brown, soft
	-880		В	8 to 16	20 to 41	Lignite, black, soft, grades down to carb. mudstone, brown, soft and then to carbonaceous claystone, dark brown
		B	С	0 to 2	0 to 5	Carbonaceous mudstone, brown, soft
-		c	D	0 to 2	0 to 5	Carbonaceous mudstone, brown, soft
			F	0 to 2	0 to 5	Thin coal stringers, black, soft, near the top and base of a
	- 870	E	L	0 to 2	0 to 5	3ft (1m) zone of weakly carbonaceous mudstone, gray-brown
2850-	870		F	10 to 18	25 to 46	Lignite, black, soft
-			G	1 to 3	2 to 5	Lignite, black, soft at/near top of 3ft (1m) carb mudstone, soft, brown
_		ر G				C.f. bentonite, carbonaceous base in places
-	-860		I	0 to 14	0 to 36	Lignite, dark brown to black, clay-rich, grades to carbonaceous mudstone at base
			J	0 to 20	0 to 20	Lignite, black, powdery, grades to carb. mudstone at base
2800		к	ĸ	32 to 42	81 to 107	Lignite, black, hard

**Figure 6.** Generalized descriptions for the upper carbonaceous beds at Tracy Mountain.



◄ Figure 7. Approximately 3.4 feet (1.04 m) of lignite is exposed after the colluvium was scraped off of measured section 6. Sample 6K2t was obtained from the top two inches (5 cm) and sample 6K2b from the basal two inches (5 cm) of the coal.



▲ Figure 8. A three-inch (7.6 cm) thick coal stringer (sample G) within a four-foot (1.2 m) thick carbonaceous mudstone at measured section no. 6.



▲ Figure 9. Sample B, measured section no. 2, was taken from the top two inches (5.1 cm) of this one-foot (0.3 m) thick organic-rich bed. As is typical of many of these thin carbonaceous beds, it consists of a mixture of thin lignite layers and even thinner lignite stringers. This bed is overlain by silty mudstone.



▲ Figure 10. A drone photograph looking southwest across the main study site at Tracy Mountain to Bullion Butte in the distance. Thick, bluish-gray swelling claystones (assumed to be bentonites) in the Sentinel Butte Formation cap many of the knolls and ridges adjacent to Tracy Mountain in this area.



▲ Figure 11. Popcorn texture of the surface of a bentonite layer resulting from the shrinking and expanding of the swelling clays as they alternately hydrate and dehydrate. A penny in the lower left for scale.



▲ Figure 12. In the foreground, a thin, carbonaceous bed (Sample O) at the base of a bentonite in measured section no. 7. Bentonite is draped over the underlying mudstone. Another prominent bentonite is present just below the break in slope. The photograph was taken looking to the northeast.



◄ Figure 13. A thin, carbonaceous bed (sample 812) beneath a bentonite (812r) in measured section 8.

A white to light gray claystone or mudstone was traceable throughout most of measured section nos. 1-9 (Figures 4 and 14). The bed ranges in appearance from a dazzling white, iron-oxide stained, wellindurated, well-exposed claystone to a light gray, moderately exposed mudstone. Wherever traceable, the one consistent trait throughout the bed was the greasy feel of the rocks which was assumed to indicate an elevated presence of kaolinite similar to what is found in the Rhame Bed of the Slope Formation (Paleocene) and the Bear Den Member of the Golden Valley Formation (Paleocene\Eocene) (Murphy, 2013).

The top of the main butte at Tracy Mountain, as well as much of the surrounding uplands, is capped by yellow to yellowish-gray siltstone, clayey siltstone, or silty mudstone (Figure 4). The clay fraction of the capstone is variable and the distinction between clayey siltstone and silty mudstone is less straightforward than the cross-section implies (Figures 15 and 16). The siltstone is typically poorly cemented, but eight of the 15 measured sections recorded flaggy, well-cemented siltstone beds within three feet (0.9 m) of the tops of buttes (Figure 17). Although there may have been some reworking of the poorly cemented siltstone by the wind, it was not evident in the measured sections along the perimeter of the buttes. Pliocene gravel is often present as lag deposits at the top of the major buttes of western North Dakota, but none was observed at Tracy Mountain nor on the adjacent uplands.

Nodules and concretions are relatively common in the Sentinel Butte Formation. Calcitecemented sandstone and siltstone concretions, along with iron-oxide nodules, were collected for analysis to investigate if critical elements were mobilized with cementing agents (Figure 18). Barite nodules are less common, well-rounded, and are typically smaller in size in comparison to the other nodules and concretions. A fist-sized barite nodule was collected while measuring section no. 7 (Figure 19).

#### Sample Results

A summary of concentrations is presented in Table 2. Individual analyses are plotted along the corresponding measured section in Appendix A and listed in a comprehensive table in Appendix B. The stratigraphic position of the rock samples was also plotted on the two cross-sections generated across the study area and color-coded based on those samples meeting or exceeding 300 ppm for total rare earth element concentrations and those concentrations falling near or below that threshold (Figures 20 and 21). In addition, the maximum and mean rare earth concentrations of samples A-K were plotted on the generalized stratigraphic section of Tracy Mountain (Figure 22). In all of these figures and tables, several patterns are evident: the highest rare earth concentrations at Tracy Mountain occur in the stratigraphically highest carbonaceous beds (A-I), high REE concentrations are sometimes found in carbonaceous zones immediately underlying bentonite beds (I and O), as well as carbonaceous beds associated with white (possibly kaolinite-rich) beds (D and E).

Two generalized targets for economic concentrations of REE in coal are referenced in the literature: 1) 300 ppm in whole coal, proposed by the U.S. Dept. of Energy, and 2) 0.1% rare earth oxide (REO) content in the ash (which excluded scandium) suggested by Seredin and Dai (2012). Both of these thresholds are referenced in this report as each examines a unique subset of samples, which is useful as it is yet unknown whether any future extraction methods will utilize feedstocks of whole coal or coal ash (see REE Composition). Many of the samples that contain over 300 ppm on a whole coal basis are rich in clay (e.g., sample 9Ir is 639 ppm REE but 92% ash), and conversely some samples that are below 300 ppm



▲ Figure 14. A bright, white, mudstone (sample 1(sil)) in measured section 1 at Tracy Mountain. The mudstone is greasy to the touch. The yellow beds cap the ridge in the background.



▲ Figure 15. A drone photograph looking northwest to the main mesa at Tracy Mountain and beds of the Sentinel Butte Formation. Yellow beds cap the mesa and dark, bluish-gray swelling claystones (bentonites) are visible along the lower one third of the exposed rocks. Square Butte and Sentinel Butte in the background.



▲ Figure 16. The capping yellow beds at Tracy Mountain consist of siltstone, silty mudstone, and silty claystone. The iron-stained rocks contain round iron-oxide spheres and iron-oxide layers.



▲ Figure 17. Flaggy siltstone caps the top of measured section no. 3 and litters the underlying slope. The thin, carbonaceous claystone (sample 3A), partially exposed in the center foreground, is high in critical elements.



▲ Figure 18. Alternating layers of claystone, mudstone, siltstone, sandstone, and lignite in the Sentinel Butte Formation are visible in this photograph of the base of measured section no. 5. A half dozen limonite nodules occur along the same approximate stratigraphic position within a swelling claystone.



▲ Figure 19. A barite nodule (left of pick) from measured section no. 7. The nodule was found lying on an ironoxide concretionary layer.

						Tracy Mou	untain Ana	alyses (co	ncentratio	ns in ppm	)	
Ch	emical Group	Flement	Symbol	Δ		Who	ole Rock Ba	asis		Ash Basis		
	ionnical creap	Liomon	- Cynnoon	~	n	MAX	MIN	MEAN	MAX	MIN	MEAN	
		Lithium	Li	3	34	76.6	1.3	29.9	92.5	4.3	42.2	
/	Alkali Metals	Rubidium	Rb	37	24	108	11	61	134	25	83	
		Cesium	Cs	55	34	9.88	0.14	4.77	12.3	0.46	6.8	
		Beryllium	Be	4	34	7.6	0.4	3.9	20.3	0.4	6.5	
	Alkaline	Magnesium	Mg	12	157	28800	1580	9020	53300	1600	14600	
E	Earth Metals	Strontium	Sr	38	34	587	34	309	1320	35	545	
	_	Barium	Ba	56	81	32000	222	3954	37800	373	6500	
		Lanthanum	La	57	169	148	4.0	35	285	6.7	56	
		Cerium	Ce	58	169	361	7.1	79	695	12.8	131	
		Praseodymium	Pr	59	72	46.7	1.3	16.8	89.2	4.3	29.5	
		Neodymium	Nd	60	169	199	3.5	43	415	6.4	73	
		Samarium	Sm	62	72	49.3	1.6	16.6	96.4	5.3	29.8	
ts:		Europium	Eu	63	72	11.5	0.53	3.9	24.7	1.45	7.1	
len		Gadolinium	Gd	64	169	42.2	0.7	9.5	103	1.2	17	
en	Lanthanides	Terbium	Tb	65	72	6.46	0.39	2.47	15.1	0.91	4.5	
Ξ	Lanthanities	Dysprosium	Dy	66	72	34.8	2.5	13.8	82.2	5.7	25.7	
arth		Holmium	Ho	67	72	6.19	0.48	2.65	15.3	1.11	5.0	
Еа		Erbium	Er	68	169	16.7	0.36	4.8	40.7	0.58	8.5	
Le		Thulium	Tm	69	72	2.21	0.19	1.01	5.37	0.49	1.92	
æ		Ytterbium	Yb	70	72	13.6	1.20	6.4	33.2	3.12	12.1	
		Lutetium	Lu	71	72	2.00	0.18	0.97	5.06	0.46	1.84	
		Scandium	Sc	21	163	41.5	0.9	17.9	120	3.1	30	
		Yttrium	Y	39	169	143	3.9	41	409	5.4	73	
	1	Titanium	Ti	22	34	6930	257	2190	7040	849	3150	
		Vanadium	V	23	126	405	5	140	732	27	211	
		Chromium	Cr	24	108	167	8	71	298	12	102	
		Manganese	Mn	25	24	1310	22	120	1740	31	170	
	Transition	Cobalt	Со	27	55	253	1.9	20	354	1.9	32	
	Metals	Zirconium	Zr	40	157	1150	10.1	310	2080	19.9	560	
	motulo	Niobium	Ni	41	126	30.8	0.8	14.2	97.1	3.2	23.3	
		Molybdenum	Мо	42	157	233	<0.2	46	402	<0.3	81	
		Hafnium	Hf	72	137	19	0.4	5	34	0.6	8	
		Tantalum	Та	73	108	2.83	0.07	0.82	3.19	0.23	1.13	
		Tungsten	W	74	154	83.5	0.1	7.7	151	0.1	13	
		Gallium	Ga	31	128	44.2	4.1	19.7	92.7	5.2	30.1	
P	ost-Transition	Indium	In	49	21	0.17	0.02	0.10	0.28	0.09	0.15	
	Metals	Tin	Sn	50	21	27	0.8	2.0	4.2	16	2.9	
	Wietais	Bismuth	Bi	83	21	1.21	0.21	0.60	2.29	0.42	0.90	
		Germanium	Ge	32	128	64	<1	10	100	0	17	
		Arsenic	As	33	73	698	17	104	1260	17	180	
	Metalloids	Antimony	Sh	51	81	25.0	0.29	7.8	44.9	0.29	14.0	
		Tellerium	Te	52	21	0.64	<0.1	0.23	0.84	<0.25	0.33	
		Thorium	Th	00	24	45.0	1.0	18.2	55.4	1.0	25.2	
	Actinides	Uranium	11	00	457	40.8	1.9	27	204	1.5	20.0	
		Oranium	0	92	157	200		21	304	1	40	

### ▼ **Table 2.** Summary of results of sample analyses from the Tracy Mountain study area.

A = atomic number

n = number of samples analyzed



▲ Figure 20. A geologic cross-section of measured sections 1-9 in the Tracy Mountain study along with the stratigraphic position of samples submitted for analysis. See Figure 3 for the location of the measured sections.



▲ Figure 21. A geologic cross-section of measured sections 1-15 in the Tracy Mountain study. See Figure 3 for the location of the measured sections.

#### Tracy Mountain Generalized Stratigraphic Column

Main butte (Measured sections 1 through 9) Upper 100 ft / 30 meters

Elevation									
ft. m	Ded	Bed th	ickness	Complex	Sample	ASH%	∑REE Conc. (ppm)		
2900	Bea	in	cm	Samples	Position	MEAN	MAX	MEAN	
A	Α	0 to 2	0 to 5	3	whole	67%	524	404	
-880	р	0 to 10	20 to 41	17	top	73%	1054	511	
	D	01010	201041	5	bottom	77%	642	473	
В	С	0 to 2	0 to 5	1	whole	85%	622	-	
- c	D	0 to 2	0 to 5	2	whole	61%	205	191	
12222222	E	0 to 2	0 to 5	4	top	75%	646	466	
-870 E	L	0 to 2	0 to 5	2	bottom	59%	782	564	
2850-		10 to 18	25 to 46	18	top	43%	402	215	
		10 10 10	251040	5	bottom	44%	282	217	
	G	1 to 3	2 to 5	9	whole	61%	412	219	
	G			6	roof	93%	639	282	
		0 to 14	0 to 36	11	top	70%	1089	681	
-860	J	01014	010 30	1	bottom	88%	413	-	
23.55.55.55	J	0 to 20	0 to 20	6	top	87%	195	163	
	ĸ	32 to 42	81 to 107	10	top	57%	134	96	
2800	Ň	52 10 42	0110107	3	bottom	20%	365	232	

▲ Figure 22. General stratigraphy and REE concentrations of the REE-enriched carbonaceous beds (samples A-K) in the upper 100 ft (30 m) at Tracy Mountain. See Appendix A for full stratigraphic columns.

contain the highest ash-basis concentrations (sample 4Kb contains only 220 ppm on a whole basis, but is low-ash at 16.5%). Of the 169 samples in this report, 53 contained whole-basis concentrations above 300 ppm REE, and 23 contained ash concentrations of 0.1% REO or greater, which here includes scandium.

There are fewer generally accepted economic thresholds for the other 28 elements investigated in this report. These elements are produced from a wide variety of ores, have supply/demand dynamics controlled by a plethora of industries, and are economic at concentrations of different orders of magnitude. In lieu of this, results can be contextualized in terms of enrichment over a baseline. A common standard for this is upper continental crust (UCC). This is not necessarily an economic proxy, however, as each element is economic at a unique ratio to UCC. The rare earths are well-known for not being "rare" in the earth's crust, but for rarely concentrating into economic deposits, thus the REE are economic at relatively low UCC multiples. Other elements naturally occur in deposits with much higher ratios and coals need to be highly enriched to be competitive, i.e., a coal that is enriched 100 fold over UCC is still not a promising source if existing ores are 1,000 times UCC. Dai and Finkelman (2018) examined concentrations of traditional ore sources vs. those reported from coals and attempted to classify the prospects for each element to be competitively produced from coal. It is therefore also relevant to contextualize concentrations of North Dakota lignites vs the average world coal (AWC). Thus the most promising critical elements would be elements that both 1) Dai and Finkelman identified as potentially competitive and 2) North Dakota lignites are especially enriched in (see Non-REE Critical Elements).

#### **REE Composition**

Although there are commercial applications for each of the rare earth elements, market prices vary significantly between them. Many of the more naturally abundant elements (e.g., cerium and lanthanum) are overproduced relative to global demand, thus the value of a given deposit depends not only on its total REE concentration, but the relative percentages of the surplus elements to those in high demand (e.g., dysprosium, neodymium, terbium). The Mountain Pass mine, the U.S.'s only major historical source of REE, targets an igneous carbonatite deposit with a rare earth composition of approximately 49.6% and 33.8% low-value cerium and lanthanum, respectively. The laterite clay deposits of southeast China, conversely, contain cerium and lanthanum values as low as 0.4% and 1.8% of total REE (Gschneidner and Pecharsky, 2019). Given the variability in compositions, the potential of two deposits that contain "economic" concentrations of 300 ppm  $\Sigma$ REE could be drastically different depending on the relative abundance of the individual elements.

A more detailed examination of the individual element distribution is thus required to assess the economic potential of a deposit, especially the behavior of the most economic REEs. The simplest classification scheme divides the REEs into heavy (dysprosium, erbium, gadolinium, holmium, lutetium, terbium, thulium, ytterbium, and yttrium) and light (cerium, europium, lanthanum, neodymium, praseodymium, samarium, and scandium) elements, although the point of division varies. The heavy elements have traditionally been considered the more valuable group, but light elements neodymium, praseodymium, and scandium have seen increased commercial use and are now among the most economically important. Seredin and Dai (2012) created a coefficient designed to illustrate the economic value of a given concentration by representing the relative abundance of elements with high demand to those produced in excess. This outlook coefficient of rare earth ores (C<sub>outl</sub>), listed below, groups the REE into critical (neodymium, europium, terbium, dysprosium, erbium, and yttrium), uncritical (gadolinium, lanthanum, praseodymium, and samarium), and excessive (cerium, holmium, thulium, ytterbium, and lutetium).

$$C_{outl} = (Nd + Eu + Tb + Dy + Er + Y / \Sigma REY)/(Ce + Ho + Tm + Yb + Lu / \Sigma REY)$$

This coefficient has been a useful qualitative proxy used by many authors since its creation in 2012, however, there have been significant changes in market conditions over the last decade as both supply and demand dynamics have evolved. Prices of lanthanum and samarium are now on par with cerium (excessive) (Table 3). Prices of yttrium, one of the more abundant elements in the critical REE group, have fallen to similar levels. Critical REE europium and erbium have also seen market prices fall below uncritical elements praseodymium and gadolinium and excessive elements lutetium and holmium which have all risen appreciably. Seredin and Dai also did not include scandium in their review, as it does not occur in most conventional REE ores, but may currently be the most valuable single rare earth element in many coals.

Table 3 also includes the theoretical maximum value of each element in one metric ton of ore based on the most enriched sample from Tracy Mountain at 2021 market prices. Although it is unlikely that ore volumes on the order of metric tons will contain concentrations on par with the maximum spot

▼ Table 3. Rare earth oxide (REO) market prices from 2011 to September 2021 and the theoretical value of one ton of ore if it were to contain the highest REE concentrations found at Tracy Mountain.

	PEO	N	<b>/</b> laı	ket Pric	ce	Max values of concentrations identified in this study											
		REO .			(US	SD/kg REO	))	W	/hole	coal basi	is		As	h basis			
		[REE] <sub>2</sub> O <sub>3</sub>		2011*		2021**	%Change	Max (REO)	V	alue/mt	%Value	Max (REO)	V	alue/mt	%Value		
		Terbium oxide	\$	557.00	\$	1,743.88	313%	7.44	\$	12.97	8.6%	17.4	\$	30.28	7.9%		
E	=	Neodymium oxide	S	50.64	\$	142.97	282%	232	\$	33.19	22.0%	484	\$	69.17	18.1%		
atic	ical	Dysprosium oxide	S	235.00	\$	454.04	193%	39.9	\$	18.13	12.0%	94.3	\$	42.82	11.2%		
ific		Erbium oxide	S	90.00	\$	54.20	-40%	19.1	\$	1.04	0.7%	46.5	\$	2.52	0.7%		
355	£	Europium oxide	S	553.00	\$	30.64	-94%	13.3	\$	0.41	0.3%	28.6	S	0.88	0.2%		
Ű		Yttrium oxide	S	143.00	\$	11.39	-92%	182	\$	2.07	1.4%	520	S	5.92	1.6%		
nic	"IE	Gadolinium oxide	S	10.71	\$	71.33	666%	48.6	\$	3.47	2.3%	119	\$	8.48	2.2%		
De la	Ξ	Praseodymium oxide	S	49.34	\$	138.25	280%	54.7	\$	7.56	5.0%	104	\$	14.43	3.8%		
8	LCI	Samarium oxide	S	16.00	\$	4.16	-74%	57.2	\$	0.24	0.2%	112	S	0.47	0.1%		
ai	2	Lanthanum oxide	S	23.82	\$	1.35	-94%	174	\$	0.23	0.2%	334	S	0.45	0.1%		
Ő		Holmium oxide	S	41.00	\$	200.31	489%	7.1	\$	1.42	0.9%	17.5	S	3.50	0.9%		
, E	ive.	Lutetium oxide	S	274.00	\$	848.38	310%	2.27	\$	1.93	1.3%	5.75	S	4.88	1.3%		
ed	ess	Ytterbium oxide	S	27.00	\$	21.21	-27%	15.5	\$	0.33	0.2%	37.8	S	0.80	0.2%		
Sel	X	Cerium oxide	S	23.10	\$	1.44	-94%	423	\$	0.61	0.4%	814	S	1.17	0.3%		
	-	Thulium oxide		-		-	-	2.52		-	-	6.13		-	-		
	Scandium oxide			-	1,060.47	-	63.7	\$	67.50	44.7%	184	\$	195.36	51.3%			

\*https://www.statista.com/statistics/449834/average-rare-earth-oxide-prices-globally/ \*\*https://www.metal.com/Rare-Earth-Oxides (Prices as of Dec 23, 2021)

concentration, the values do illustrate that even with full recovery, the economic potential of a REE deposit is largely controlled by a handful of the REE (scandium, neodymium, dysprosium, terbium, and perhaps praseodymium).

Another informative look at composition involves correlating the individual elements to total rare earth concentrations, i.e., do all elements undergo linear enrichment or are some more relatively abundant in the more enriched samples? Sixty-six samples in the Tracy Mountain study area were analyzed for all 16 rare earth elements, including scandium and yttrium. While this sample subset is biased toward enriched samples, and thus not representative of the average lignite in the study area, it does offer an opportunity to examine compositional change across a more evenly distributed range of SREE concentrations from 62 to 1,089 ppm (whole coal basis). Increased REE fractionation is observable in more enriched samples, represented graphically in Figure 23 and numerically as r in Table 4. Compositional trends in Tracy Mountain lignites show preferential enrichment in the light REEs. Note that heavy REE concentrations also increase as total REE concentrations increase, but not to the same rate of lighter REEs, e.g., although scandium increases from an average of 13.7 to 35.0 ppm between the bottom ten and top ten SREE samples, its average as a percentage of SREE decreases from 9.0% to 4.2%, a strong negative correlation. Conversely, cerium has a moderately strong positive correlation, increasing from 23.5% to 31.9% across the same samples. These trends have implications for modeling recovery of the most economic elements. As bulk ore quality increases (higher ΣREE concentrations), proportionally less of that total will consist of scandium, dysprosium and terbium, while neodymium and praseodymium will make up larger proportions.



▲ Figure 23. Compositional trends in the REEs across a subset of 66 samples with analyses of all 16 REE. Elements in green make up a larger proportion of total rare earths as the samples become more enriched, while elements in red show the opposite trend.

▼ Table 4. Compositional data for the subset of 66 samples with analyses of all 16 REE. The light REEs make up a larger proportion of the total rare earth concentrations the more a sample is enriched.

	Floment		Conce	ntrations	(ppm)	Compostistion (% of total REE)								
	Liement		MAX	MEAN	MIN	MAX	MEAN	MIN	r %∑RE	E				
	Cerium	Ce	361	135	9.9	37.1%	28.8%	16.0%	0.55	▲				
ъ	Praseodymium	Pr	46.7	17.8	1.3	4.8%	3.8%	2.1%	0.44					
R	Neodymium	Nd	199	75	5.7	22.4%	16.3%	9.2%	0.32	u of				
ght	Lanthanum	La	148	56	5.2	17.7%	12.1%	6.2%	0.31	rat				
	Samarium	Sm	49.3	17.5	1.6	5.8%	3.8%	2.2%	0.19	tich				
	Europium	Eu	11.5	4.1	0.53	1.3%	0.9%	0.6%	0.16	E, G				
	Gadolinium	Gd	42.2	16.9	2.1	5.4%	3.8%	2.5%	-0.05	т of				
	Terbium	Tb	6.46	2.60	0.39	0.8%	0.6%	0.4%	-0.19	ate me				
s	Dysprosium	Dy	34.8	14.5	2.5	4.5%	3.3%	1.8%	-0.38	rich I				
Ш	Yttrium	Y	143	67	11.8	29.7%	15.9%	7.9%	-0.43	e Co				
Ž	Holmium	Ho	6.19	2.78	0.48	1.0%	0.7%	0.3%	-0.47					
leav	Erbium	Er	16.7	7.8	1.36	3.1%	1.9%	0.9%	-0.52					
-	Thulium	Tm	2.21	1.06	0.19	0.5%	0.3%	0.1% -0.56						
	Ytterbium	Yb	13.6	6.7	1.20	3.3%	1.6%	0.8%	-0.58	*				
	Lutetium	Lu	2.00	1.01	0.18	0.5%	0.2%	0.1%	-0.58					
	Scandium	Sc	41.5	24.3	5.9	19.2%	6.1%	3.2%	-0.62					

7' is the correlation coefficient (from +1 to -1)

#### **Non-REE Critical Elements**

Regardless of the final economic outlook for producing rare earths alone from coal, its prospects could increase substantially if additional valuable minerals were present that could be co-produced from the same coal. Several mineral commodities are already commercially produced from coal or coal ash, including aluminum, gallium, germanium, magnesium, silicon, selenium, and vanadium (Dai and Finkelman, 2018). Extraction methods are being developed for additional highly promising elements like gold, silver, platinum, palladium, molybdenum, niobium, rhenium, and zirconium. Highly elevated concentrations of antimony, beryllium, chromium, cesium, iridium, iron, hafnium, lithium, osmium, rhodium, ruthenium, tantalum, titanium, and tungsten have been found in coal or coal ash that could be competitive with conventional ores and are potentially one supply/demand disruption away from receiving commercial interest.

Some concentrations of these elements in North Dakota lignites have been reported, as summarized in the USGS's COALQUAL database (Palmer et al., 2015; Lin et al., 2018b). Many critical elements are strongly represented with as many as 7,600 analyses reported, but less than 200 represent North Dakota coals. With so few samples, it is not surprising none of the critical element concentrations in North Dakota's lignites are among the top in the COALQUAL database, which is primarily bituminous Appalachian coals. These 200 entries also likely do not reflect the state's most mineralized seams, instead being full-bed samples of the thick, low-ash coals preferred for supplying thermal power plants. Furthermore, representing the full thickness of a major lignite with one mixed concentrations is likely to dilute any signal from enriched zones at the top or bottom of the bed, if present. Concentrations from thin lignites at Tracy Mountain represent new highs for essentially all of the elements reported from North Dakota coals and are higher than any in the national COALQUAL database for magnesium, molybdenum, vanadium, zirconium and many of the REEs (terbium, dysprosium, holmium, samarium, erbium, thulium, ytterbium, and scandium) (Lin and others 2018a, 2018b). Gallium and uranium concentrations reported here are also within 10% of the highest in the database (Figure 24).

In their review of elemental concentrations reported from coal, Dai and Finkelman (2018) classified the potential of each element to be produced from coal based on its competitiveness with currently utilized ore sources for each element. The elements enriched at Tracy Mountain (vs the most enriched US coals reported in the COALQUAL database) fall in the most favorable categories: already commercially produced from coal (gallium, magnesium, vanadium), historical production from coal (uranium), or enriched coals are very promising to be competitive with traditional ores (REE, molybdenum, zirconium).

#### **Organic Association**

Another consideration when identifying mineral commodities to potentially co-produce with REEs is the inferred organic affinity of the element. Coals with a low ash yield have been considered the favored targets for potential REE extraction, not only because burning these coals can concentrate non-combustible mineral matter twenty-fold or more, but because this coal type is already utilized for thermal power generation, thus offering opportunity for an additional revenue stream utilizing existing mining



▲ Figure 24. The highest concentrations at Tracy Mountain (21 to 157 analyses per element) compared to the highest concentration in the USGS COALQUAL database (often over 7,500 analyses per element). Note: higher concentrations of magnesium, titanium, and barium from natural coal ash, bentonite, and a barite nodule, respectively, from Tracy Mountain are not included.

infrastructure. REE extraction from coal ash can be problematic however, as acid-leaching studies have found that a portion of the REE becomes immobilized in glassy matrix during burning (Taggart and others 2016). REE recovery therefore could be more effective on whole coal in some basins, in which case the more desirable feedstocks may include clay-rich coals, partings, seam roofs and floors, which often have higher whole-basis REE content and a separate group of associated minerals. A large portion of the REE in North Dakota lignites have been shown to be organically associated and easily recovered from precombustion coal feedstocks (Laudal et al., 2018).

Organic beds at Tracy Mountain are relatively clayey: 43 of 53 samples >300 ppm  $\Sigma$ REE on a whole basis contain ash yields over 50% by weight. Lithologies of these samples are predominantly coaly stringers in dark brown carbonaceous mudstone (B and E beds) or black clayey coal (I bed) and carbonaceous lenses in its bentonite roof (Ir samples). Despite the higher clay content of beds B, E, and I, six, two, and six samples from these beds, respectively, exceed 0.1% REO on an ash basis. Samples >300 ppm  $\Sigma$ REE and <50% ash yield occur in the more coal-rich F and G beds, but whole basis REE concentrations are generally lower and only two ash samples from each bed exceed 0.1% REO.

Using the ash yield as a general proxy for clay content, many critical elements reported in this study are associated with inorganic fractions of the sample and are unlikely candidates for REE coproduction in low ash lignites in North Dakota. The elements titanium, cesium, and lithium have a very strong positive correlation with the ash yield, represented as *r* ASH% in Table 5. Also strongly correlated are gallium, rubidium, tin, tantalum, and thorium (a contaminant). None of these elements with strong or very strong inorganic associations are current targets of production nor were they considered highly promising by Dai and Finkelman (2018), with the exception of gallium. Of the other more promising elements, magnesium and vanadium had moderate positive correlations, niobium had a weak positive correlation with the ash yield. The inorganic associations inferred for samples at Tracy Mountain are largely consistent with other lignites from North Dakota, although zirconium and niobium show a stronger positive correlation to ash yield across other sites (Moxness, 2020).

The rare earth elements show mostly weak correlations to ash yield at Tracy Mountain. This is to be expected, as the occurrence of moderate to high REE concentrations in low-ash coals (that become high to very high REE concentrations in coal ash) is the primary reason the rare earths in coal have attracted recent interest in the U.S. as opposed to clays or other siliciclastic deposits. Elements with similarly low inorganic affinities are not necessarily enriched via the same pathways as the rare earths, however, which is a primary consideration in identifying elements for co-production. Column  $r \Sigma$ REE in Table 5 indicates the correlation coefficient between each element and the total REE. The individual REEs, intuitively, exhibit a very strong positive correlation to  $\Sigma$ REE. Highly promising elements vanadium and gallium (moderately strong) molybdenum, zirconium, and niobium (weak), and germanium and magnesium (very weak) all show at least some positive correlation to coals enriched in rare earths.

#### **Enrichment Models**

Identifying the mechanism(s) by which lignites become enriched in rare earths and other critical elements is an important step in modeling the geologic context of potential ore sources in North Dakota. Seredin and Dai (2012) recognized four main genetic types of REE-enriched coals: terrigenous, tuffaceous, infiltrational, and hydrothermal. Terrigenous and tuffaceous enrichment are proposed to have occurred during deposition, where REE were transported to the peat bog in surface water, often in association with the leaching of tuffaceous sediment. Infiltrational enrichment occurs post-deposition via descending groundwater flows and is consistent with the model for uranium enrichment in coals in North Dakota identified by Denson and Gill (1965). Hydrothermal enrichment is associated with ascending fluids transporting REE into a coal at any stage of its development.

	Correlatio	n with	total	REE		Correlation	with	ash y	ield
	Element		n	r ∑RE	E	Element		n	r
	Cerium	Се	70	0.99		Zirconium	Zr	143	-0.
	Praseodymium	Pr	70	0.98		Molybdenum	Мо	143	-0.
	Neodymium	Nd	70	0.97		Germanium	Ge	115	-0.
	Terbium	Tb	70	0.97		Tungsten	w	140	-0.
	Dysprosium	Dy	70	0.96		Hafnium	Hf	125	0.0
ente	Gadolinium	Gd	70	0.96		Strontium	Sr	31	0.0
ē	Europium	Eu	70	0.96		Arsenic	As	67	0.0
Ē	Holmium	Но	70	0.95		Manganese	Mn	21	0.1
art	Samarium	Sm	70	0.95		Cobalt	Со	51	0.1
ш е	Lanthanum	La	70	0.95	•	Uranium	U	143	0.1
Ra	Erbium	Er	70	0.93		Lutetium (REE)	Lu	66	0.1
	Thulium	Tm	70	0.92		Ytterbium (REE)	Yb	66	0.1
	Ytterbium	Yb	70	0.91		Thulium (REE)	Tm	66	0.1
	Yttrium	Y	70	0.89		Erbium (REE)	Er	66	0.2
	Lutetium	Lu	70	0.87		Antimony	Sb	75	0.2
	Scandium	Sc	64	0.84	Ш	Barium	Ba	75	0.2
	Uranium	U	143	0.69	ΣR	Holmium (REE)	Но	66	0.2
	Indium	In	21	0.67	vith	Niobium	Nb	113	0.2
	Chromium	Cr	97	0.63	nt v	Dysprosium (REE)	Dy	66	0.2
	Beryllium	Ве	31	0.63	me	Gadolinium (REE)	Gd	66	0.2
	Antimony	Sb	75	0.61	rich	Terbium (REE)	Тb	66	0.2
	Vanadium	V	113	0.60	-en	Samarium (REE)	Sm	66	0.2
	Gallium	Ga	115	0.54	о Б	Yttrium (REE)	Y	66	0.2
	Titanium	Ti	31	0.54	sin	Europium (REE)	Eu	66	0.2
	Thorium	Th	21	0.50	crea	Tellerium	Те	21	0.2
	Lithium	Li	31	0.50	Ĕ	Neodymium (REE)	Nd	66	0.3
	Cesium	Cs	31	0.48		Indium	In	21	0.3
	Hafnium	Hf	125	0.48		Praseodymium (REE)	Pr	66	0.3
	Molybdenum	Мо	143	0.41		Cerium (REE)	Се	66	0.4
	Zirconium	Zr	143	0.40		Magnesium	Mg	143	0.4
	Tellerium	Те	21	0.29		Scandium (REE)	Sc	64	0.4
	Niobium	Nb	113	0.29		Bismuth	Bi	21	0.4
	Tungsten	w	140	0.24		Vanadium	V	113	0.4
	Arsenic	As	67	0.22		Chromium	Cr	97	0.5
	Bismuth	Bi	21	0.18		Beryllium	Ве	31	0.5
	Barium	Ba	75	0.17		Lanthanum (REE)	La	66	0.5
	Tantalum	Та	97	0.14		Thorium	Th	21	0.0
	Magnesium	Mg	143	0.13		Gallium	Ga	115	0.0
	Cobalt	Со	51	0.13		Tantalum	Та	97	0.7
	Germanium	Ge	115	0.11		Tin	Sn	21	0.7
	Tin	Sn	21	0.09		Rubidium	Rb	21	0.1
	Rubidium	Rb	21	-0.02		Lithium	Li	31	0.9
	Strontium	Sr	31	-0.09		Cesium	Cs	31	0.9
	Manganese	Mn	21	-0.29		Titanium	Ti	31	0.9

▼ Table 5. The likelihood of each critical element to be enriched with the REE (left) and whether each element is more enriched in low- or high-ash lignites (right) at Tracy Mountain.

> r ASH% -0.14 -0.10 -0.04 -0.02 0.00 0.01 0.06 0.10 0.10

coals

in low-ash

enrichment

0.15 0.16

0.17 0.19 0.20

0.20 0.21 0.21

0.21 0.21

0.23

0.23

0.25

0.25

0.25

0.28

0.31

0.33 0.36

0.41

0.41 0.42 0.47 0.47

0.51 0.52 0.53 0.63 0.69 0.72 0.77

0.78 0.90 0.93 0.96

.⊑

enrichment

Relative

¥



n is the number of analyses

7' is the correlation coefficient (from +1 to -1)

Seredin and Dai (2012) also described three REE distribution patterns (L-type (light), M-type (medium), and H-type (heavy)) in high-REE coals based on the preferential transport and sorption of certain element groups corresponding to the different genetic types of enrichment. These patterns are classified using the ratios of individual REE concentrations normalized to upper continental crust ([REE]<sub>N</sub>) (Taylor and McLennan, 1985). In this study, 70 of 169 samples were analyzed for all of the lanthanide elements (excluding promethium) and yttrium, including all 53 samples projected to contain concentrations over 300 ppm on a whole basis (see Laboratory Analysis). Twenty-three samples meet the Seredin and Dai (2012) criteria for high rare earth accumulation in coal of 0.1% REO (rare earth oxide) content in the ash, which here includes scandium.

Seven of the 53 enriched samples in this study classify as L-type ( $La_N/Lu_N > 1$ ), all from the I bed. The I bed is a clayey carbonaceous bed up to 14 inches (36 cm) in thickness and is overlain by a prominent bentonite (weathered volcanic ash). This is consistent with the description of L-type distributions that are a product of tuffaceous enrichment via volcanic ash input into the peat. The seven L-type samples are clay-rich portions of the I bed (76% to 96% ash), including two roof samples of carbonaceous lenses in bentonites. The REE concentrations of these samples range from 394 to 691 ppm. One sample reaches 0.1% REO in the ash, although less-clayey areas of this bed contain the highest REE concentrations in the Tracy Mountain study area and lower  $La_N/Lu_N$  ratios.

The most enriched samples in this study, including 19 of 23 samples over 0.1% REO on an ash basis, are consistent with M-type distribution patterns (Fig. 4B), with  $La_N/Sm_N < 1$  and  $Gd_N/Lu_N > 1$  and a consistent  $Eu_N$ -maximum (Figure 25). This enrichment pattern was also attributed to natural acidic water interacting with tuffaceous sediment during the peat bog stage, and the higher concentrations of the medium-REEs is likely due to higher humic matter in these samples relative to the clayier L-type samples.

Three of the 23 samples over 0.1% ash REO and three of the 53 samples over 300 ppm REE contain concentrations of  $La_N/Lu_N < 1$  and  $Gd_N/Lu_N < 1$ , which classify as H-type. Heavy enrichment can occur in any of the genetic types where more alkaline waters transport REE in terrigenous, tuffaceous, infiltrational, or hydrothermal settings.

In addition to identifying light, medium, and heavy distributions, which are each associated with multiple genetic types, Seredin and Dai also used associated non-rare earth elements as an additional line of evidence in determining the source of enrichment of high-REE coals: terrigenous (aluminum, gallium, barium, strontium), tuffaceous (zirconium, hafnium, niobium, tantalum, gallium), infiltrational (uranium, molybdenum, selenium, rhenium), and hydrothermal (arsenic, antimony, mercury, silver, gold).

Although the presence of bentonite and predominately medium-light fractionation of the REE suggest tuffaceous enrichment, samples from Tracy Mountain do not contain high normalized concentrations of zirconium, hafnium, niobium, tantalum, or gallium. The ash samples most enriched with these elements contain concentrations 3.2 ( $Ta_N$ ), 4.5 ( $Ga_N$ ), 5.9 ( $Hf_N$ ), 8.1 ( $Nb_N$ ), and 10.9 ( $Zr_N$ ) times that of the upper continental crust (UCC) (Table 6). Some elements associated with the hydrothermal genetic type, by contrast, are abundant. Antimony in the ash is up to 225 times UCC and arsenic is up to 840 times UCC. Elements associated with the infiltration model are also considerably more abundant. The highest normalized uranium concentrations approach 109 times UCC and molybdenum is up to 268 times UCC.



▲ Figure 25. Enrichment in Tracy Mountain samples (ash basis) relative to upper continental crust. Samples variably contain relative enrichment in light (blue), medium (green), or heavy (red) REE.

A more targeted examination of co-enriched critical elements involves comparing concentrations from Tracy Mountain lignites to the average values for world coal (Ketris and Yudovich 2009). Among coal ash samples >0.1% REO in this study, arsenic and antimony (associated with hydrothermal enrichment) average only 4.0 and 3.5 times the average world coal (AWC), while uranium and molybdenum (associated with infiltrational enrichment) average 6.3 and 10.7 times AWC. At Tracy Mountain, the concentrations of several critical elements (antimony, arsenic, barium, germanium, molybdenum, and uranium) average over 5X the upper continental crust in coal and 10X on an ash basis (Table 6), although this does not necessarily indicate they are enriched in the same pathways as the REE. In the 23 most REE-enriched coal ashes at Tracy Mountain, average concentrations of beryllium, bismuth, lithium, manganese, rubidium, strontium, tantalum, tin, and titanium are still below the average world coal ash. In the clayier whole coal subset of 53 REE-enriched samples, only bismuth and manganese average concentrations are below the average world coal.

		Enrichment coefficients vs Upper Continental Crust (UCC)														Enrichment coefficients vs Average World Coal (AWC)								
			All Tr	acy Mo	untain li	ionite sa	moles			REE-en	riched 1	racy M	ountain	lignite s	amples		AWC REE-enriched Tracy Mountain lignite sample						oles	
	lice		Whole	coal ha	sis		Ash hasi	8	>30	0 nnm Σ	REE (w	hole)		0 1% R	EO (ast	h)	(01	nm)	>300 ppm ΣRFF (whole)			>0.1% REO (ash)		
Flowerst	(ppm)	-	MAX	MEAN		Luny.	MEAN			MAY	ME AN			MAX			b.ala		ΜΑΥ ΜΕΛΝ ΜΙΝ					
Lanthanum	30	155	MAX	1.2	0.1	0.5	1 Q	0.4	51	MAA	2.2	0.8	23	0.5	MEAN 4.7	2.6	whole 11	asn 69	13.5	6.0	2.2	MAA 4.1	2.0	1 1
Cerium	64	155	5.6	1.2	0.1	10.9	2.1	0.4	51	5.6	2.2	1.3	23	10.9	5.7	3.2	23	130	15.5	6.8	3.6	5.3	2.0	1.1
Drasendymiur	71	70	6.6	2.3	0.1	12.6	4.2	0.4	51	6.6	2.0	1.3	23	12.6	6.9	3.8	35	20	13.7	5.9	2.7	4.5	2.0	1.0
Neodymium	26	155	77	17	0.2	16.0	3.0	0.5	51	7.7	3.4	1.0	23	16.0	8.2	4.5	12	67	16.6	7.3	3.0	6.2	3.2	1.7
Ø Samarium	4.5	70	11.0	37	0.4	21.4	6.7	12	51	11.0	4.5	1.6	23	21.4	11.3	5.8	20	13	24.7	10.2	3.7	7.4	3.9	2.0
E Europium	0.88	70	13.1	44	0.4	28.0	8.1	1.6	51	13.1	5.5	2.3	23	28.0	13.6	6.5	0.47	25	24.5	10.2	4.3	9.9	4.8	2.0
E Gadolinium	3.8	155	11.1	2.6	0.0	27.2	47	0.5	51	11.1	5.2	2.6	23	27.2	13.0	7.5	27	16	15.6	73	3.7	6.4	3.1	1.8
Terbium	0.64	70	10.1	3.9	0.6	23.6	7.1	1.4	51	10.1	47	2.4	23	23.6	11.8	7.1	0.32	21	20.2	9.4	4.8	72	3.6	22
E Dyspros	3.5	70	9.9	4.0	0.0	23.5	73	1.6	51	9.9	4.8	2.6	23	23.5	12.1	7.0	21	14	16.6	8.0	4.3	5.9	3.0	17
ш Yttrium	22	155	6.5	1.0	0.2	18.6	3.3	0.2	51	6.5	3.5	1.9	23	18.6	8.5	47	84	51	17.0	9.1	4.8	8.0	37	2.0
Holmium	0.8	70	77	3.3	0.6	19.1	6.2	1.4	51	77	4.0	22	23	19.1	10.0	5.7	0.54	40	11.5	5.9	3.3	3.8	2.0	11
C Erbium	23	155	73	2.2	0.0	17.7	3.7	0.3	51	7.3	3.9	21	23	17.7	9.7	5.4	0.93	5.5	18.0	9.6	5.2	74	4.1	2.3
Thulium	0.33	70	6.7	3.1	0.6	16.3	5.8	1.5	51	67	3.7	2.0	23	16.3	9.2	4.9	0.31	2.0	7.1	3.9	2.1	27	1.5	0.8
Ytterbium	22	70	6.2	2.9	0.5	15.1	5.5	1.4	51	6.2	3.4	1.8	23	15.1	87	4.3	10	62	13.6	7.6	4.0	5.4	3.1	1.5
Lutetium	0.32	70	6.3	3.1	0.6	15.8	5.8	1.5	51	6.3	3.6	1.8	23	15.8	9.0	4.1	0.20	12	10.0	57	2.9	42	2.4	11
Scandium	13.6	149	3.1	1.3	0.0	8.8	22	0.2	50	3.1	2.0	0.8	22	8.8	4.5	27	3.9	23	10.6	6.9	2.0	52	2.6	1.6
Antimony	0.2	75	125	40.3	1.5	225	73.6	9.1	27	125	54.2	4.5	13	225	109	19.6	0.92	6.3	27.2	11.8	1.0	7.1	3.5	0.6
Arsenic	1.5	67	465	71.8	3.4	840	127	4.5	25	309	87.0	82	12	323	126	31.1	83	47	55.9	15.7	1.5	10.3	4.0	1.0
Barium	550	75	31.8	6.5	0.4	68.6	11.5	0.7	27	25.1	8.5	0.6	13	55.4	16.5	1.2	150	940	92.0	31.0	2.1	32.4	9.6	0.7
Bervllium	3	31	2.5	1.4	0.4	6.8	2.3	1.1	19	2.5	1.6	1.0	8	3.6	3.0	1.7	1.6	9.4	4.8	3.0	1.8	1.2	0.9	0.6
Bismuth	0.127	21	9.5	4.9	1.7	18.0	7.1	3.3	17	9.5	5.1	2.3	7	18.0	7.9	4.8	0.97	5.9	1.2	0.7	0.3	0.4	0.2	0.1
Cesium	4.6	31	2.1	1.0	0.0	2.7	1.5	0.1	19	2.1	1.4	0.2	8	2.3	1.5	0.6	1.0	6.6	9.9	6.2	0.9	1.6	1.1	0.4
Chromium	83	97	2.0	0.8	0.1	3.6	1.3	0.2	39	2.0	1.1	0.3	16	3.6	2.0	1.0	16	100	10.4	5.9	1.7	3.0	1.7	0.8
Cobalt	17	51	14.9	1.2	0.1	20.8	2.0	0.2	23	14.9	1.5	0.4	10	5.0	2.2	0.5	5.1	32	49.6	5.0	1.4	2.7	1.2	0.3
Gallium	17	115	2.6	1.2	0.2	5.5	1.8	0.5	42	2.6	1.5	0.5	19	5.5	2.7	1.3	5.8	33	7.6	4.3	1.4	2.8	1.4	0.7
German	1.6	115	40.0	6.7	0.6	62.7	11.5	1.3	42	26.3	6.6	1.9	19	45.6	16.3	4.1	2.2	15	19.1	4.8	1.4	4.9	1.7	0.4
Hafnium	5.8	125	3.3	0.9	0.1	5.9	1.5	0.4	45	2.6	1.2	0.5	19	3.7	2.4	0.7	1.2	8.3	12.5	5.6	2.4	2.6	1.7	0.5
Indium	0.05	20	3.4	2.1	1.0	5.7	3.1	1.7	17	3.4	2.3	1.4	7	5.7	4.1	2.5	0.031	0.16	5.5	3.7	2.3	1.8	1.3	0.8
Lithium	20	31	3.8	1.5	0.1	4.6	2.2	0.2	19	3.8	2.1	0.3	8	4.3	2.5	0.8	12	66	6.4	3.4	0.5	1.3	0.8	0.2
Magnesium	13300	143	1.8	0.7	0.1	4.0	1.1	0.2	48	1.8	0.7	0.2	22	4.0	1.4	0.5	n/a	n/a	-	-	-	-	-	-
Manganese	600	21	2.2	0.2	0.0	2.9	0.3	0.1	17	0.4	0.1	0.0	7	0.3	0.2	0.1	86	490	2.7	0.8	0.3	0.3	0.2	0.1
Molybdenum	1.5	143	155	32.3	0.7	268	58.0	0.8	48	155	47.3	4.8	22	232	99.9	12.4	2.2	14	106	32.3	3.3	24.8	10.7	1.3
Niobium	12	113	2.6	1.2	0.1	8.1	2.0	0.4	41	2.6	1.4	0.4	18	6.5	3.0	1.0	3.7	20	8.3	4.4	1.4	3.9	1.8	0.6
Rubidium	112	21	1.0	0.6	0.1	1.2	0.8	0.2	17	1.0	0.6	0.1	7	0.9	0.6	0.2	14	79	7.7	4.6	0.8	1.3	0.9	0.3
Strontium	350	31	1.7	0.9	0.2	3.8	1.6	0.3	19	1.6	0.9	0.2	8	3.1	1.6	0.7	110	740	5.0	2.8	0.8	1.5	0.8	0.3
Tantalum	1	97	2.8	0.8	0.1	3.2	1.2	0.2	39	1.6	0.9	0.2	16	2.2	1.2	0.4	0.28	1.7	5.8	3.3	0.6	1.3	0.7	0.3
Tellerium	n/a	21	-	-		-	-	-	-	-	-	-	-	-	-	-	n/a	n/a	-	-	-	-	-	-
Thorium	10.7	21	4.3	1.9	0.3	5.1	2.6	0.7	17	4.3	2.1	1.1	7	4.1	3.2	2.4	3.3	21	13.9	6.7	3.5	2.1	1.6	1.2
Tin	5.5	21	0.5	0.4	0.1	0.8	0.5	0.3	17	0.5	0.4	0.2	7	0.8	0.6	0.4	1.1	6.4	2.5	1.9	1.0	0.7	0.5	0.4
Titanium	4100	31	0.9	0.5	0.1	1.0	0.7	0.2	19	0.9	0.6	0.1	8	1.0	0.8	0.3	800	4650	4.5	3.2	0.7	0.9	0.7	0.3
Tungsten	2	140	41.8	3.9	0.3	75.4	7.0	0.5	47	18.3	5.4	0.8	21	23.0	9.0	1.4	1.1	6.9	33.3	9.9	1.5	6.7	2.6	0.4
Uranium	2.8	143	71.4	10.2	0.4	109	16.4	1.6	48	71.4	19.9	1.8	22	109	36.1	11.3	2.4	16	83.3	23.2	2.1	19.0	6.3	2.0
Vanadium	107	113	3.8	1.3	0.0	6.8	2.1	0.3	41	3.8	1.9	0.2	18	6.8	3.4	0.7	25	155	16.2	7.9	0.7	4.7	2.3	0.5
Zirconium	190	143	6.1	1.7	0.1	10.9	3.2	0.3	48	6.1	2.1	0.7	22	10.9	5.6	0.9	36	210	31.9	11.0	3.4	9.9	5.1	0.9

▼ Table 6. Tracy Mountain elemental concentrations normalized to upper continental crust and world coals.

UCC values from Taylor and McLennan (1985) and McLennan (2001); AWC values from Ketris and Yudovich (2009) n is number of samples analyzed

#### Conclusions

Tracy Mountain is noteworthy for its consistent lateral rare earth element enrichment across multiple thin lignite beds. Although these beds are too thin to be economically mined for their critical element contents, they provide new insights into the variability and mineral associations within REE-enriched lignites in North Dakota and the geologic context in which these deposits occur. The upper perimeter (roughly 4,500 ft or 1,400 m) of the main butte of Tracy Mountain contains multiple thin (~1 ft) lignites that are consistently enriched in REE (over 300 ppm, up to 1,000 ppm whole coal basis). The area encompassed by this butte is roughly 25 acres (100,000 square meters), suggesting high REE concentrations in North Dakota lignites can occur over a more widespread area than would be expected if enrichment were the product of localized vertical groundwater flows.

High REE concentrations are found in carbonaceous beds immediately underlying bentonite beds (I and O), as well as carbonaceous beds associated with white (possibly kaolinite-rich) beds (D and E), suggesting enrichment occurred from volcanic input directly into the peat or the weathering of adjacent

volcanic sediments, perhaps as soon as immediately after during the Paleocene. Although this enrichment model is consistent with tuffaceous enrichment proposed by Seredin and Dai (2012), high to very high REE concentrations also occur in stratigraphically higher beds (A, B, and C) with no immediate proximity to bentonites or the white (weathered) zone. Furthermore, lignites below bentonites lower in section do not show similarly high enrichment. Enrichment in the upper beds may have occurred later as overlying volcanogenic sediment of the White River Group was eroded post-Miocene, or it may reflect Quaternary leaching from the weathering of a long-lived upland surface.

This upland proximity hypothesis was tested by sampling the stratigraphically highest carbonaceous beds in the areas adjacent to the badlands topography. Uppermost carbonaceous beds on nearby buttes were often not enriched, and no samples approached the 1000 ppm REE seen in multiple beds on the main butte of Tracy Mountain. REE enrichment in the uppermost sample is not uncommon from other sites in North Dakota (Kruger et al, 2017), but input from Quaternary weathering is likely not a factor in other exceptionally enriched lignites like the H bed in Slope County, which has hundreds of feet of overburden, including 25 or more feet of other non-enriched coals, where it contains REE concentrations in excess of 1000 ppm (Murphy et al., 2018). Results from the Tracy Mountain area, as well as the authors' previous reports suggest multiple pathways of rare earth enrichment in North Dakota lignites.

The bulk of the market value of a REE deposit comes from a small number of the 16 rare earth elements. In the 53 samples over 300 ppm total REE at Tracy Mountain, the four most economic elements (dysprosium, neodymium, scandium, and terbium) make up 25.7% of the total REE, a significant improvement over the 15.3% in the U.S.'s traditional source of REE, Mountain Pass Mine. Other economically important elements are also enriched in these lignites. Spot concentrations of gallium, magnesium, molybdenum, vanadium, uranium, and zirconium at Tracy Mountain are nearly as high or higher than any beds in the USGS COALQUAL national database. These elements could thus be potential candidates for co-production with REE if similar co-enrichment trends extend to other North Dakota lignites. At Tracy Mountain, gallium, vanadium, and uranium especially tend to be enriched in the same samples as the REEs. The most promising of the elements investigated are generally considered to be gallium, germanium, magnesium, molybdenum, niobium, vanadium, uranium, and zirconium, but the ultimate prospects for each will depend not only on bulk concentration but on the ease of co-production by newly developed REE extraction technologies.

Ongoing research is examining the REE recovery economics from raw clay-rich coal vs coal ash. At Tracy Mountain, elevated concentrations of the more promising elements gallium, magnesium, scandium, and vanadium correlate with the ash yield, suggesting these elements may be more abundant in clay-rich lignites. Conversely, molybdenum, niobium, germanium, uranium, zirconium and many of the heavy REE have very weak to negative correlation with the ash yield, suggesting more organic association and higher potential from low-ash coals. The rare earth elements have been shown to have multiple modes of occurrence in North Dakota lignites, with the heavy REEs associated with organic complexes in the coal and the lighter REEs with mineral matter (UND-IES, 2021). Titanium, cesium, lithium, and rubidium are very strongly correlated to ash yield. Establishing the presence of these elements in North Dakota lignites and the tendency for each to co-occur has implications in identifying the feedstocks, target mineral commodities, and extraction methods for any future commercial production.

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# Appendices

Appendix A

Legend for lithologies of measured sections

Carbonaceous Claystone/Mudstone
White Claystone/Mudstone
Lignite
••• Nodules and Concretions
Covered



								Lab A	nalysi	is (in j	ug/g)							
									_	m							To Ri	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	Praseodymi	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
1B1 1B2	191 124	18.8 11.1	10.1 6.01	5.37 3.16	21.8 13.0	3.64 2.14	83.1 57.6	1.33 0.79	97.7 61.7	24.1 15.1	21.9 13.3	30.6 22.3	3.34 1.96	1.38 0.83	8.72 5.37	88.7 53.2	612 392	955 492
1F 1(sil)	38.8 41.1		3.03 1.28		6.3 2.1		15.2 22.4		25.3 16.6			18.2 6.2				24.6 10.7	156 112	336 114
1ir 1i	52.9 116	21.1	2.29 13.0	4.68	4.4 21.9	4.48	29.1 58.9	1.73	24.4 70.3	15.7	17.3	15.2 32.7	3.46	1.74	10.9	20.5 112	169 506	178 879
1K	18.9		1.81		2.6		8.1		10.2			4.9				20.0	78	209

												Lab	Analy	vsis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
1B1 1B2	7.33		4240	3.1		6.33	76 84		22.9 23.8	7 10	6.3 5.4		37.6	8610 11100		78.0 47.8	15.5 13.8		420	0.83 1.02				2740	13.3 10.0	65.5 70.6	165 156	358 288
1F 1(sil)	1.44	2.8	367	0.5		5.29	55 36	1.9	10.1 10.2	15 2	8.7 2.8		33.3	8010 1580	38	67.9 1.4	15.0 26.1	49	34	0.33 1.78		6.8		6930	9.9 3.7	27.1 3.1	57 54	739 97.7
1lr 11	1.99	16.1	3680	1.9		6.62	88 157	12.5	21.3 24.7	7 42	2.5 10		32.8	9090 6400	157	4.5 94.4	13.4 26.8	96	173	0.97 0.67		9.8		4310	2.2 3.4	4.1 27.9	138 405	89.0 1150
1K							8		5.5	3	2.9			7550		16.5	4.5			0.16					7.9	6.3	17	298



								Lab A	nalysi	s (in p	ug/g)							
									_	m							To RE	tal E
ple ID	m	orosium	m	pium	olinium	nium	hanum	tium	dymiur	eodym	arium	ndium	ium	ium	rbium	m	le coal	
Sam	Ceri	Dysl	Erbi	Euro	Gad	Holr	Lant	Lute	Neo	Pras	Sam	Scar	Terb	Thul	Ytte	Yttri	Who	Ash
2A	61.5		5.80		8.6		31.8		32.6			22.8				43.0	243	324
2B1t 2B2 2B1b	203 361 205	18.1 28.5 18.8	10.2 15.2 10.4	5.94 8.67 5.04	23.1 35.4 21.6	3.52 5.4 3.67	91.9 148 97.6	1.44 1.97 1.31	112 175 98.9	26.4 42.4 24.4	24.6 38.1 21.4	31.0 41.5 39.9	3.37 5.22 3.36	1.46 2.06 1.41	9.36 13.4 8.83	76.1 132 80.0	641 1054 642	862 1508 798
2Ft 2Fb 2G 2I	50.3 62.4 34.1 301	28.4	4.82 4.38 3.86 15.5	9.42	8.9 8.3 4.0 34.0	5.47	19.1 24.0 17.5 107	1.91	32.5 35.3 15.7 172	40.1	40.7	15.2 18.0 12.0 30.9	5.00	2.11	13.6	37.9 34.1 33.8 130	204 221 141 937	492 407 211 1683
2K	31.6	2.9	1.73	0.75	3.1	0.57	13.9	0.26	15.6	4.0	3.4	6.9	0.48	0.25	1.70	14.0	101	225
2N	9.9	2.6	1.95	0.53	2.1	0.61	5.2	0.34	5.7	1.3	1.6	11.9	0.39	0.30	2.04	15.5	62	205
2Q	55.0	6.6	3.94	1.82	7.5	1.32	21.0	0.58	33.3	7.7	7.8	12.9	1.14	0.56	3.79	31.1	196	409
2R	20.2	2.5	1.36	0.79	3.0	0.48	7.9	0.18	13.5	3.1	3.3	5.9	0.44	0.19	1.20	11.8	76	232

												Lab	Analy	/sis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
2A	22.4	392	11700	5.4	0.65	6.61	81	100	25.2	14	10	0.07	26.5	18600	1310	139	28.4	87	587	0.74	0.15	11.9	2.0	2420	56.6	18.9	184	480
2B1t 2B2 2B1b	12.8 9.39 6.9	329 285 89.6	9830 11500 1390	5.0 7.6 5.6	0.59 0.86 0.65	5.38 5.77 9.88	89 110 118	9.5 9.5 9.8	32.6 44.2 34.5	13 15 20	13 15 8.2	0.11 0.14 0.12	56.4 60.4 54.7	11400 15000 11300	56 22 35	233 159 75.5	22.0 29.1 19.5	41 48 108	523 548 353	1.23 1.54 1.05	0.42 0.59 0.20	21.6 30.7 27.9	2.4 2.5 2.3	2540 2830 3250	24.6 31.2 18.3	100 157 106	241 201 206	775 959 503
2Ft 2Fb 2G 2I	9.59 6.49 6.22 25.0	266 48.4 43.4 270	1650 486 1340 581	2.9	0.34	4.38	102	10.8 5.7 12.7 16.5	17.5	29	6.9 5.3 5.9 6.2	0.07	20.7	6030 4780 11800 5310	86	78.5 42.2 35.7 58.5	13.4	45	255	0.56	0.11	14.2	1.4	1960	15.2 5.1 2.8 5.7	23.1 18.6 26.9 42.4	160	541 292 299 533
2К	4.88		517	1.9		1.33	19		10.5	10	3.3		9.7	3710		12.5	9.7		320	0.70				900	1.9	6.0	51	219
2N	6.95		775	1.9		0.14	11		6.3	11	2.4		1.3	4260		22.1	9.0		399	0.07				257	4.3	9.1	24	295
2Q	10.5		978	1.9		3.36	44		9.8	14	3.0		7.7	3010		68.3	10.9		204	0.32				1140	9.1	19.4	93	392
2R	6.11		986	1.1		0.87	25		4.3	14	1.9		3.4	1970		11.0	16.8		279	0.18				747	4.2	4.6	68	217



									Lab A	nalysi	is (in J	ug/g)							
											Ξ							To	tal FF
	Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	Praseodymiu	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
	34	533		9 01		10.9		26.8		30.2			32.7				72 7	283	512
4	3B1t 3B2t 3B3 3B4	141 122 73.3 90.6	14.7 10.8 9.1	7.95 6.16 5.13 5.53	4.27 3.15 2.68	17.4 12.4 8.5 9.9	2.78 2.10 1.86	53.2 55.3 33.2 42.0	1.06 0.85 0.79	81.0 60.4 38.3 44.3	19.2 15.1 10.9	18.9 13.2 9.6	28.2 26.4 19.2 21.7	2.66 1.91 1.55	1.10 0.86 0.78	7.00 5.52 5.12	64.8 51.4 40.6 49.4	465 388 255 306	1077 487 <i>435</i> 416
	3B1m	68.0		4.03		7.5		31.1		34.8			16.7				35.9	230	332
	3B1b 3B2b	122 128	13.3 13.3	7.72 7.91	3.47 3.64	14.8 14.3	2.64 2.68	58.2 62.8	1.05 1.10	62.3 63.7	15.2 15.6	14.2 13.6	26.7 31.5	2.33 2.26	1.06 1.08	6.80 7.03	67.6 65.1	419 434	731 555
	3Ft 3Fb	34.9 29.8		4.70 3.75		6.5 4.7		15.9 14.5		20.6 14.9			14.1 17.4				41.7 32.7	166 139	452 338
	3G	46.8		4.34		5.3		23.9		21.6			15.8				38.5	182	265
	311 312	314 161	30.0 15.5	13.1 7.10	10.9 5.33	38.6 19.1	5.05 2.72	95.1 63.6	1.44 0.83	183 91.2	41.7 21.7	46.5 22.5	40.7 26.1	5.84 2.98	1.70 0.96	10.4 5.80	101 59.8	939 506	1678 612
	3J	51.8		1.91		3.4		26.9		21.8			18.0				15.6	156	173
	ЗК	20.3		0.83		1.5		10.2		8.9			8.7				6.8	64	86

												Lab	Analy	ysis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
34	16.4	698	16800					57			19			23200		126									83 5	44 4		870
3B1t 3B2t 3B3 3B4	6.17 8.66 6.03 6.28	87.0 186 175 86.2	4860 6280 1590 7370	4.4 4.6 4.0 4.2	0.29 0.75 0.45 0.54	1.37 7.65 4.80 8.97	72 79 54 80	10.9 12.2 26 17.6	17.3 30.8 21.9 32	6 9 7 8	7.1 7.4 6.1 6.3	0.11 0.09 0.05 0.07	15.4 58.7 29.0 38.1	10300 9100 9700 10300	46 43 39 113	48.1 85.8 56.1 87.8	16.2 15.9 14.7 16.4	11 64 37 69	471 268 252 299	0.45 1.47 1.09 1.16	0.18 0.26 0.16 0.13	12.4 18.9 10.0 13.6	1.1 2.5 1.7 2.0	1270 2790 1900 3000	10.8 11.0 9.8 13.4	60.7 74.9 114 76.0	107 185 117 181	334 389 297 283
3B1m	5.11	30.8	2780	3.1	0.72	6.37	50	6.3	31.7	6	4.8	0.06	46.0	5870	30	41.0	10.9	56	242	1.94	0.13	15.9	2.7	2380	8.4	32.7	126	193
3B1b 3B2b	9.18 13.1	42.0 267	1540 13800	3.8	0.55	5.52	90	16.3 9.8	22.9	11	6.2 7.6	0.09	18.2	6470 8130	29	62.8 107	11.5	66	482	0.61	0.23	11.5	1.4	2150	10.4 10.9	64.9 94.8	227	323 494
3Ft 3Fb	7.06 6.18	48.6 33.4	1720 1580					8.4 7.7			6.5 4.7			11300 9170		75.2 44.1									12.7 7.1	24.5 12.1		418 284
3G	8.10	34.9	3210					30.9			5.2			15900		37.9									2.3	22.7		266
311 312	19.3	65.6	376	5.4	0.35	5.97	167 129	23	25.5 29.6	23 5	6.2 3.3	0.12	29.5	5210 5310	86	64.0 8.7	9.7 11.4	56	199	0.57 0.85	0.14	22.4	1.6	2190	2.3 2.3	81.4 22.6	285 202	489 180
3J							90		26.1	2	3.4			11000		2.3	11.6			0.94					2.0	4.5	150	124
зк							21		22.2	3	3.8			7010		1.5	3.8			1.26					0.7	6.2	37	96.1



								Lab A	nalys	is (in J	ug/g)							
										Ē							To R	tal EE
₽		sium		Ε	ium	ε	m	F	nium	lymiu	Ę	ε	_	_	Ξ		oal	
mple	erium	/spro:	bium	Iropiu	adolin	olmiu	nthar	tetiu	sodyr	aseoc	mari	andiu	rbiun	uliun	terbiu	trium	hole c	٩
Sa	Ű	á	Ъ	EL	Ö	Ť	La	3	ž	Pr	Sa	Sc	P_∎	È	≯	¥	>	As
4B1t 4B2	134 127	12.7 14.0	7.18 7.94	3.43 3.63	14.5 15.4	2.48 2.80	56.1 60.2	0.96 1.04	69.2 64.4	17.4 15.8	15.1 14.7	28.9 23.4	2.22 2.41	0.99 1.08	6.23 6.87	64.6 65.0	436 426	505 607
4B1b	137	15.1	8.97	4.00	16.5	3.06	66.6	1.09	71.9	17.4	15.9	29.0	2.58	1.17	7.13	81.2	479	549
4F1t	46.0		3.56		6.1	4.60	22.3	0.67	25.5	6.2		14.5	4.20	0.00	4.27	29.4	173	271
4F2	44.7	7.8	4.65	1.64	7.6 6.9	1.60	17.2	0.67	26.4 25.0	0.2	0.0	13.4	1.28	0.00	4.27	29.6	178	427 392
4G1	34.3		4.82		4.9		19.0		16.0			13.1				46.2	162	249
4G2 4I1	32.7	20.6	4.44 9.51	6 4 1	4.5 24 3	3 62	15.8 96.9	1 00	14.9	28.4	26.4	14.4 37.0	3.85	1 23	7 22	38.3 88.5	147 691	249 905
412	216	20.0	8.91	6.29	24.3	3.49	97.0	0.94	111	27.4	26.4	32.4	3.82	1.15	6.75	84.8	671	818
4J	52.6		2.04		4.0		25.7		23.6			17.0				16.5	160	175
4Kt 4Kb	32.0 65.7	8.5	1.87 4.94	2.15	3.5 8.9	1.69	14.4 19.1	0.70	17.3 34.5	7.9	8.3	12.2 13.7	1.45	0.69	4.45	16.2 37.3	113 221	<i>186</i> 1338

												Lab	Analy	vsis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
4B1t 4B2 4B1b	13.7 11.9	148 142	747 11200	6.7	0.88	8.67	95	11.2 12	31.8	8	6.5 6.7	0.14	62.4	9990 10400	58	125 74.0	20.9	65	275	1.46	0.64	27.1	2.7	3620	20.4 16.7	69.3 73.8	248	278 271
4F1t 4F2 4F1b	8.13 7.04 8.47	152 93.2	1860 4180 553	3.4		1.47	15	10.2 8.5	12.2	6	7.5 8.4 6.0		5.4	9780 9420 8840		62.3 99.8 108	3.6		434	0.17				506	14.9 19.3 16.4	26.6 21.6 34.7	58	506 617 401
4G1 4G2	12.0 5.35	52.3 31.1	9280 6980					30.6 9.8			7.3 7.0			9040 17800		32.6 11.3									6.3 3.0	13.0 23.0		395 397
411 412	7.65 8.13	35.6 18.1	5560 509	6.2 5.4	0.58 0.65	6.75 7.53	118 136	7.1 8.4	23.6 26.7	5 5	2.9 3.2	0.16 0.17	39.6 49.1	5380 5190	69 63	14.2 13.4	9.5 10.8	68 81	190 85	0.65 0.77	0.10 0.12	29.6 25.7	1.8 2.0	2810 3390	2.2 2.2	44.5 34.4	198 205	137 168
4J	2.18	11.5	937					6.1			3.3			9790		4.4									1.7	5.9		123
4Kt 4Kb	4.16 5.37	37.3 12.7	845 2730					3.7 11.5			5.0 3.2			6540 2490		7.0 12.6									2.4 4.8	7.4 5.2		285 241



					_			Lab A	Analys	is (in <sub>l</sub>	ug/g)							
		_							E	ium							To R	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymiun	Praseodym	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
5Bt 5Bb	251 121	20.3 10.9	10.5 6.24	6.01 2.99	25.2 12.3	3.72 2.16	112 61.5	1.27 0.76	130 59.1	32.5 14.9	28.3 12.4	32.9 21.2	3.79 1.95	1.39 0.84	8.47 5.08	89.0 57.0	756 390	935 463
5E 5F1 5F2	109 42.6 46.6	13.5 9.8	6.70 3.90 5.05	4.23 2.81	17.3 9.6 12.3	2.46 1.82	40.5 13.4 14.1	0.87 0.66	79.6 40.1 48.7	17.1 9.1	19.8 12.8	23.2 14.3 19.8	2.59 1.79	0.89 0.68	5.87 4.50	49.3 30.3 36.8	393 <i>189</i> 227	503 <i>452</i> 549
5G 5Ir 5It 5Im 5Ib	125 50.6 141 79.8 142	14.6 11.3 10.0	8.57 1.88 5.54 2.38 4.87	2.86 3.36 3.10	14.5 3.6 13.6 6.0 12.2	2.94 2.04 1.80	52.5 27.4 67.6 38.8 68.6	1.17 0.65 0.58	57.8 21.7 65.1 35.3 65.5	14.4 16.6 16.8	12.9 14.4 13.6	19.3 19.5 23.6 12.0 21.3	2.44 2.09 1.87	1.17 0.73 0.66	7.64 4.38 4.03	74.1 16.6 52.7 21.9 45.7	412 158 425 223 413	952 172 500 234 467
5К	29.9		1.77		3.7		13.7		16.5			7.5				16.8	105	216
5L	37.5		2.39		3.3		19.6		17.2			16.2				20.5	133	176
5(nod)	11.8		0.62		1.2		8.1		5.7			2.9				7.2	43	62

# Tracy Mtn. REE Section 5 T.135N., R.101W., Sec.10, NW/SE/NW Elevation at top 2,886 ft.

												Lab	Analy	/sis (in	µg/g	)								_				
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
5Bt 5Bb							86 71		28.6 30.5	98	7.5 4.8			14000 11700		142 86.7	18.2 13.3			1.24 1.42					18.5 11.6	99.8 111	194 189	297 228
5E 5F1 5F2							88 36		26.5 8.3	7 4	6.3 6.0			12300 6150		29.8 89.7	15.0 9.7			1.03 0.28					3.0 21.1	31.4 17.4	213 63	345 379
5G 5Ir 5It 5Im 5Ib							53 116 117 66 100		17.3 28.4 35.1 17.8 25.3	4 3 5 3 4	2.9 3.0 2.3 3.1			8930 6430 4500 7510		6.5 13.0 4.3 7.2	10.0 14.2 12.6 11.5 13.2			0.74 0.91 0.83 0.83 0.90					2.5 2.7 1.5 2.3	15.5 3.8 11.7 4.4 10.6	207 209 107 167	201 98.2 136 79.4 124
5К							24		12.8	8	4.4			5880		7.8	12.2			0.91					2.7	5.6	50	281
5L							98		24.1	28	4.8			18200		45.9	23.9			0.74					2.0	19.6	219	477
5(nod)							98		24.1	28	4.8			18200		45.9	23.9			0.74					2.0	19.6	219	477



								Lab A	nalys	is (in µ	ug/g)							
									_	mn							To Ri	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymiun	Praseodymi	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
6B1 6B2 6B3	143 105 159	13.2 11.1 17.4	7.65 6.32 10.5	3.74 2.96 4.36	14.9 12.0 18.6	2.64 2.19 3.60	65.9 47.9 78.1	1.05 0.86 1.39	69.1 53.3 77.6	17.0 13.2 19.0	15 11.7 17	30.0 24.4 30.9	2.29 1.83 2.97	1.07 0.88 1.43	6.59 5.61 8.98	67.0 54.8 105	460 354 556	522 438 759
6E 6F1 6F2	165 19.6 47.2	20.0 4.5 10.8	9.70 2.99 7.53	6.26 0.86 1.72	25.9 3.9 9.0	3.58 0.98 2.44	59.8 7.9 19.5	1.18 0.42 1.09	119 11.6 26.7	27.0 2.7 6.3	29.5 3.1 6.8	28.2 9.7 19.6	3.86 0.68 1.60	1.28 0.42 1.06	8.16 2.66 6.76	75.9 28.9 71.0	584 101 239	891 202 746
6F3 6G	83.7 114	12.7 13.7	6.58 8.32	3.24 2.63	14.6 13.2	2.37 2.78	30.1 54.6	0.82 1.15	57.9 53.9	12.9 13.6	14.2 12.1	19.4 21.0	2.28 2.26	0.87 1.16	5.47 7.47	57.8 73.6	325 395	727 856
6lr 6l	61.9 84.0		2.37 3.41		5.0 6.5		32.1 41.5		28.5 36.5			15.0 23.2				21.2 28.1	189 254	203 279
6J 6K1 6K2t 5K2b	49.4 19.6 35.9 102	13.6	2.30 1.03 1.91 7.52	3.82	4.1 1.8 3.7 15.0	2.62	24.0 9.5 17.9 39.1	1.03	22.6 9.5 18.5 61.7	14.1	15.3	18.6 7.6 11.6 25.8	2.39	1.06	6.74	18.4 8.8 14.1 52.9	158 66 120 365	186 112 178 1551

												Lab	Analy	sis (in I	µg/g)							_						
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
6B1 6B2 6B3	10.7	93.8	5740	4.5	0.87	8.32	122 87 70	25.8	36.9 22.3 32.1	13 7 7	6.3 6.4 7.8	0.09	58.3	10600 15000 13000	112	105 64.7 178	20.1 17.2 17.4	96	370	1.18 1.63 1.26	0.24	21.1	2.6	3000	20.3 16.5 36.1	69.4 63.9 131	268 200 196	302 331 442
6E 6F1 6F2	15.4 7.78 9.29	67.9 113 54.9	547 3050 600	3.7 6.5		0.64 1.19	13 23	32 8.3 29.9	10.7 17.5	6 6	8.7 4.7 6.5		6.1 8.7	11900 8550 9600		51.3 55.8 92.6	4.0 5.5		318 400	0.12 0.19				436 596	4.5 8.9 10.0	49.2 14.9 34.7	59 80	524 297 426
6F3	4 50	34.4	1850				27	16.7	8.9	5	5.9 6.8			7030		77.2 26.1	5.2			0.20					7.3	16.5 23.7	47	385
6lr 6l	4.50	54.4	2000				84 106	10.7	21.1 28.4	3 5	2.8 4.5			20500 12500		1.7 6.4	14.6 13.9			1.03 1.03					2.1 2.2	2.9 15.6	128 202	90.6 254
6J 6K1 6K2t 6K2b	6.80 4.30	23.5 37.8	3940 7510					5.9 2	18.4 21.8	7 11	4.2 3.2			13500 6730 6630 4280		11.2 8.3 9.8 9.8									1.9 1.1	12.3 5.0 10.3 8.0		214 158 204 487



								Lab A	nalysi	is (in µ	ug/g)							
										m							To RI	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	Praseodymi	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
7B	151	14.6	8.35	4.11	16.9	2.89	69.8	1.16	76.8	18.7	17.2	28.0	2.59	1.17	7.43	79.8	501	605
7D 7E1t 7E1b 7E2 7F1 7F2t 7F3 7F4b 7G1 7G2 7J1 7J2 7Kt	40.2 183 91.9 216 84.8 114 44.6 69.9 72.2 56.8 38.9 45.8 65.0 26.7 35.3	7.4 25.0 11.4 30.8 17.6 12.4 17.3 7.5 10.3 3.3 3.4 4.0	4.74 12.0 5.65 13.99 6.38 8.81 1.99 6.38 8.73 4.20 6.51 5.89 5.19 3.67 2.20 2.35 1.92 2.26	1.41 6.96 3.80 9.85 4.75 3.55 4.67 1.88 2.24 0.84 0.86 1.14	7.0 30.4 15.0 21.0 3.2 15.4 20.0 6.7 4.4 3.5 4.6 4.3	1.58 4.54 2.04 5.29 3.24 2.30 1.45 2.09 0.70 0.66 0.80	18.4 58.8 31.7 61.1 28.7 6.7 23.7 36.2 17.0 29.6 28.6 31.3 22.0 23.5 33.8 12.7 15.3	0.70 1.44 0.75 1.71 1.13 0.81 1.13 0.81 0.81 0.81 0.37	23.6 131 72.2 175 81.8 12.7 62.6 80.1 33.1 37.0 39.3 23.6 16.0 21.3 28.0 15.9 18.8	5.4 28.8 15.3 37.6 16.8 13.5 17.7 7.0 9.6 5.3 3.9 4.5	5.9 31.3 18.2 45.5 21.1 16.2 20.1 8.3 9.3 3.8 3.8 4.4	12.6 23.0 27.1 33.3 20.9 7.4 19.3 15.7 23.6 18.9 13.5 10.3 19.3	1.16 4.70 2.21 5.79 3.15 2.25 2.93 1.27 1.71 0.51 0.54 0.65	0.67 1.58 0.76 1.84 1.17 0.59 0.59 0.35 0.35	4.33 9.83 4.98 11.5 7.48 5.49 7.43 3.82 5.29 2.35 1.85 2.02	41.3 94.1 42.9 93.0 68.8 17.9 45.3 61.7 55.0 48.9 34.4 17.0 20.0 15.0 17.0	176 646 346 782 81 315 399 282 272 217 151 131 195 91 111	652 1022 523 1480 1083 <i>181</i> 786 695 498 <i>343</i> <i>221</i> <i>220</i> 154
7(bar) 7M 7N	12.6 57.9 19.3	5.2	0.57 3.34 3.95	0.88	1.2 5.4 3.7	1.23	6.6 28.1 9.6	0.64	6.3 26.9 10.4	2.4	2.7	3.1 19.2	0.68	0.61	3.93	5.3 25.1 34.0	41 191 99	41 277 338
70r 70 7P 7Q	70.8 99.6 48.6 31.7	10.1	2.80 5.45 3.19 3.26	2.89	5.7 11.9 6.5 3.2	1.91	37.4 44.1 19.3 17.6	0.75	32.8 54.6 29.7 14.5	12.9	12.4	13.4 10.5 14.1 16.4	1.83	0.76	4.84	25.7 45.1 25.3 26.2	215 320 173 130	225 344 332 221
7R	28.5	3.0	1.47	1.13	3.8	0.55	11.4	0.18	18.2	4.1	4.4		0.53	0.20	1.25	12.0	91	
75	54.8		2.77		5.7		23.4		28.8			16.9				22.0	179	250
7HTm1 7HTm2 7HTm3	21.5 9.3 7.1		1.64 0.47 0.36		2.8 0.8 0.7		12.8 6.4 4.0		11.5 3.9 3.5			5.0 1.8 2.4				18.2 6.0 3.9	85 32 25	171 107 142

												Lab	Analy	vsis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
7B	10.1	164	5630	6.6	0.91	7.62	99	14.5	33	8	7.4	0.11	76.6	8300	33	96.7	20.9	82	322	1.54	0.30	23.8	2.6	2990	20.7	58.8	214	359
7D 7E1t 7E1b 7E2 7F1 7F2t 7F3 7F4t	22.6 11.2	107	4480 11300	5.6 3.6	1.21	4.08 0.94	89 107 84 30 18	45	34 18.0 15.3 8.1 5.3	16 6 14 10 6	10 7.3 11 6.6 4.6	0.15	23.2 5.9	12200 10800 9430 7790 9040	43	124 51.9 131 110 144	30.8 14.5 25.8 6.4 6.5	46	163 203	0.80 0.70 0.56 0.16 0.15	0.32	16.0	2.2	1800 520	12.6 3.8 6.4 17.1 10.3	34.9 26.5 79.2 25.0 7.3	382 238 387 74 40	720 404 839 372 324
7F5 7F2b 7F4b							38		13.8	4	4.2			7970		33.5	13.7			0.88					3.8	19.7	92	199
7G1 7G2							<b>49</b> 47		<b>11.8</b> 12.7	<b>7</b> 7	<b>4.7</b> 4.5			<b>9030</b> 6300		<b>28.5</b> 31.4	<b>20.7</b> 17.9			<b>0.82</b> 1.05					<b>3.1</b> 3.1	<b>7.7</b> 6.4	<b>156</b> 140	<b>212</b> 193
7J1 7J2 7Kt 7Kb							89		26.9	3	3.6			12600		1.7	13.3			0.96					1.9	6.9	147	154
7(bar) 7M 7N	0.29	1.7	32000	0.4		2.70	12 88	2.3	5.1 20.3	1 46	0.6 4.0		7.8	3150 7390	84	0.5	3.2 20.4	28	524	0.24		1.9		930	0.5	1.7 15.0	27 200	19.6 360
70r 70 7P 7Q							78 56 49 65		20.0 16.0 11.1 19.9	3 4 25 59	3.3 3.1 3.5 4.1			13500 6580 4910 6950		2.3 8.5 66.7 24.8	15.5 13.8 18.1 29.9			0.89 0.84 0.39 0.53					1.5 1.6 11.4 3.4	2.9 5.1 17.6 14.7	121 103 97 171	135 237 634 469
7R																												
75							81		19.4	36	2.8			7860		18.8	21.4			0.59					1.5	16.6	211	237
7HTm1 7HTm2 7HTm3	1.94 0.55 0.46	36.7 11.2 5.1	2030 1620 1780	1.6	0.21	1.71	23	6.1 2 3.1	6.3	4	1.3 0.8 0.7	0.02	11.9	4840 4580 3800	118	6.0 4.8 2.3	5.5	28	330	0.36	<0.1	3.5	0.8	1230	1.6 1.1 1.1	4.6 1.7 1.1	35	63.8 28.4 25.4

# Tracy Mtn. REE Section 8 T.138N., R.101W., Sec.1, NE/NW/NE Elevation at top 2,889 ft.



								Lab A	nalys	is (in I	µg/g)							
										um							To Ri	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	Praseodymi	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
8(fe)	20.9		1.54		2.4		10.9		10.3			7.7				14.1	78	88
8A	120	19.5	13.1	4.25	19.2	4.25	61.6	2.00	66.3	15.2	15.9	37.5	3.14	1.86	12.3	128	524	732
8B	101	9.1	5.35	2.73	11.1	1.84	46.6	0.78	51.7	12.5	11.1	20.2	1.65	0.74	4.77	49.5	331	438
8C	163	21.0	12.5	4.76	23.4	4.45	79.7	1.50	84.1	20.0	17.6	32.4	3.60	1.60	9.38	143	622	732
8(crb)	66.2		2.78		5.3		36.0		29.1			12.1				29.4	205	215
80	74.4		4 03		6.5		39.6		34.1			17.8				35.3	242	265
0.0	10.5		4.00		0.5	4 -0		0.70	04.1			17.0	4.50			33.5	242	200
8F1 8F2	42.5 30.5	8.8	4.89 3.63	2.24	10.0 6.0	1.70	14.5	0.70	36.7 21.9	7.5	9.9	17.1	1.52	0.69	4.59	34.0 31.8	197	300
8G 8l1r 8l2r 8l1 8l2	33.7 46.5 146 189 352	7.1 16.5 34.8	4.64 1.65 3.58 7.35 16.7	2.80 5.52 11.5	4.6 3.4 9.9 20.7 42.2	1.28 2.82 6.19	21.6 24.3 69.8 78.0 129	0.43 0.79 1.91	14.6 20.3 67.1 101 199	17.4 24.4 46.7	12.9 24.2 49.3	11.2 19.4 19.2 30.6 35.0	1.42 3.18 6.46	0.47 0.95 2.21	3.03 5.82 13.4	47.5 13.8 31.1 65.3 143	161 145 394 576 1089	241 158 425 692 1848
8J	56.8		2.41		4.1		28.5		24.8			18.8				19.9	175	217
8К	37.7		2.52		4.7		17.0		20.4			9.9				21.8	134	251

												Lab	Analy	sis (in	µg/g)													
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
8(fe)	13.7	152	614					20.6			1.2			2650		31.3									2.3	22.9		49.9
8A	15.9	464	6020					253			12			23400		39.3									36.6	17.6		599
8B 8C 8(crb)	5.59 14.0 3.15	146 32.8 19.0	4940 724 618	3.0	0.46	5.02	74	7.1 15.2 37.2	20.9	8	7.4 7.3 3.0	0.07	39.4	12500 13000 8000	230	48.8 8.5 2.9	22.2	60	360	1.06	0.20	14.4	1.9	2820	20.0 3.1 1.9	44.3 27.2 8.4	150	337 545 116
8D 8F1 8F2 8G 8l1r 8l2 8l1 8l2 8J 8J 8K	7.38 6.74 6.88 4.31 7.95	34.4 291 23.9 23.4	6920 3580 17500 6200 541	1.9	0.62	0.51 8.42	29 104 117 103 122 96 34	19.6 11.8 12.9 7.9	6.4 26 24.9 28.2 33.5 28.1 11.6	4 2 6 4 20 3 11	6.6 5.6 4.7 2.6 3.3 2.5 4.5 3.6 2.9	0.12	2.6 40.4	17400 8400 7870 7100 6450 6390 5240 11500 5270	56	11.7 73.1 53.2 18.9 14.1 2.9 47.0 2.8 8.7	5.3 12.2 11.6 12.8 9.7 11.6 6.4	87	132	0.15 0.78 0.88 0.83 0.53 0.96	0.11	45.9	2.2	599	2.6 14.6 10.0 2.1 2.5 2.2 2.2 1.9 1.2	12.1 19.9 10.4 8.8 5.7 39.7 4.6 60.4 8.4 4.6	55 181 209 172 230 162 72	371 345 332 208 80.0 167 77.1 391 181 194



								Lab A	nalysi	is (in p	ug/g)							
									_	m							To Ri	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymiun	Praseodymi	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
9В	237	24.3	13.3	6.55	27.5	4.71	100	1.74	122	29.9	27.7	31.3	4.28	1.79	11.4	111	754	1148
9F	44.4		4.49		7.5		17.5		27.2			15.9				39.1	187	419
9ir 9i	226 268	14.8 31.7	7.47 15.7	5.32 9.52	19.1 39.8	2.69 5.89	101 110	0.90 1.95	111 160	28.2 34.4	23.4 37.2	20.4 34.1	2.84 5.82	1.00 2.07	6.28 12.8	68.8 125	639 894	695 2317
9К	26.4		1.42		2.7		11.9		13.6			8.0				11.6	87	139

												Lab	Analy	/sis (in	µg/g)													
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
98							85		19.8	10	8.1			12300		175	20.2			0.85					22.2	200	155	413
9F							35		16.0	9	9.7			4590		69.1	10.4			0.23					14.9	26.3	99	741
9lr 9l							117 98		32.8 21.9	8 15	2.9 6.7			5960 2360		9.2 134	13.3 18.2			0.94 0.26					2.4 4.7	10.4 51.3	189 216	110 697
9К							25		14.3	7	2.9			6090		3.1	4.1			0.67					0.6	4.5	62	117



								Lab A	nalysi	is (in j	ug/g)							
		-			_				۶	ium							To Ri	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymiur	Praseodym	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
10(crb)	54.0		2.69		4.9		23.2		24.4			17.5				21.8	171	213
10A 10B	103 101	11.2 9.3	5.69 5.21	3.29 2.59	13.6 10.2	2.04 1.78	35.0 43.6	0.74 0.72	66.6 51.1	15.3 12.7	15 11.1	15.9 22.6	2.08 1.62	0.78 0.72	5.12 4.61	40.7 45.8	336 325	377 431
10F	51.3		3.03		5.8		23.2		27.6			13.3				26.4	175	285
10G	88.5		4.33		7.3		43.6		39.7			13.9				37.0	269	303
10J	49.1		1.91		3.2		24.8		20.9			17.7				15.2	149	163
10K	32.1		1.21		2.6		16.5		15.4			11.3				10.2	101	149

# Tracy Mtn. REE Section 10

		_						_				Lab	Analy	ysis (in	µg/g													
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
<b>10(crb</b> )							59		20.6	12	9.8			20500		71.6	23.1			0.88					30.1	25.0	103	539
10A 10B							80 113		27.1 25.3	13 4	5.8 4.6			17400 7750		13.6 27.9	21.3 14.3			0.96 1.05					5.9 3.9	22.0 46.6	236 249	280 222
10F							55		16.4	12	5.6			7760		73.9	8.0			0.54					6.2	28.4	133	422
10G							48		33.8	4	5.7			5860		21.2	29.5			2.83					7.7	12.1	95	235
10J							85		26.5	3	3.6			10700		6.3	12.5			1.08					2.2	7.0	145	148
10K							49		18.8	14	3.7			5940		7.7	13.0			1.22					2.0	6.6	120	229

# Tracy Mtn. REE Section 11



[									Lab /	Analys	is (in J	ug/g)							
											Ę							To R	tal EE
	ple ID	ш	orosium	ш	pium	olinium	nium	hanum	tium	dymium	eodymii	arium	dium	ium	ium	rbium	m	le coal	
	Sam	Ceri	Dysp	Erbi	Euro	Gad	Holr	Lant	Lute	Neo	Pras	Sam	Scan	Terb	Thul	Ytte	Yttri	Who	Ash
	44.54	10.6		2.22		2.5		25.4		20.4			40.7					454	172
	11B1 11B2	49.6		2.32 1.94		3.5 3.7		25.1		20.1			13.5				14.4 16.6	151 146	172
	11C	74.6		2.81		6.2		35.0		34.6			10.8				25.3	217	230
	11J	39.9		1.76		2.8		20.2		17.4			11.5				14.3	122	137
	11K	19.9		1.04		2.2		9.1		11.2			9.5				8.6	71	103

												Lab	Analy	vsis (in	µg/g)													
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
11B1 11B2							108 73		24.5 25.8	3 2	5.3 3.0			11600 7600		33.6 6.3	14.0 9.3			1.27 1.22					3.6 1.9	14.1 7.7	233 147	258 132
11C							50		17.8	2	2.7			8020		1.1	14.1			1.20					2.6	4.5	74	99.5
11J							71		16.4	7	3.0			10200		1.5	11.2			0.87					1.5	5.5	118	210
11K							25		15.8	11	3.0			6090		9.7	5.4			0.89					2.0	6.3	48	186

![](_page_58_Figure_0.jpeg)

								Lab A	nalys	is (in p	ıg/g)							
		F			c		_		E	nium							To Ri	tal EE
nple ID	ium	sprosiur	ium	mnido.	doliniun	lmium	ithanum	etium	odymiu	seodym	narium	ndium	bium	milu	erbium	rium	ole coal	
San	Cer	Dys	Erb	Eur	Gae	ЮН	Lan	Lut	Nec	Pra	San	Sca	Ter	Thu	Ytte	Yttı	ЧХ	Ash
12E 12F	31.2 43.4		1.89 1.37		2.4 2.6		16.8 22.0		13.3 18.9			10.5 9.3				16.2 11.1	105 122	161 136
12G	28.8		1.14		2.1		12.6		12.2			7.5				11.0	85	120
12L	40.4		2.44		4.2		19.4		21.2			13.4				17.9	138	195
12P	83.3		4.90		9.3		40.0		42.3			16.4				39.6	275	335
12R	20.6		1.42		3.4		7.2		15.8			4.7				12.9	79	251

												Lab	Anal	ysis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
12E 12F 12G							67 56 13		14.7 17.2 14.0	15 3 4	3.2 2.9 2.8			11300 8350 11600		11.3 2.4 11.7	13.5 13.1 3.1			0.70 0.98 1.03					2.2 2.5 1.1	49.3 8.0 29.0	228 109 54	287 117 85.9
12L							73		22.3	64	5.0			6230		105	18.4			0.66					8.3	16.3	195	593
12P							97		22.6	30	3.5			7830		17.4	15.1			0.74					1.3	15.2	281	387
12R							29		4.5	6	2.3			2320		17.6	27.4			0.26					5.9	3.6	44	300

![](_page_60_Figure_0.jpeg)

								Lab A	nalys	is (in µ	ug/g)							
		_							5	ium							To R	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymiun	Praseodym	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
13J1 13J2	26.4 39.1		2.02 2.84		2.4 3.8		13.9 21.1 28.8		11.1 17.5 24.6			13.2 16.7				17.6 27.8	99 147 169	112 174
13L	60.0		2.32		4.3		31.4		26.5			14.0				18.3	177	182
13N	41.4		1.83		2.8		22.7		18.1			10.3				15.3	127	146
13U 13P	78.3		3.34		6.5		41.5		14.b 35.9			11.5				29.5	243	203
13Rt 13Rm	39.4 17.1		2.03 1.09		4.9 2.2		17.0 7.4		25.4 10.5			13.8 5.3				15.9 9.3	138 62	276 158

												Lab	Anal	/sis (in	µg/g	)	1		1			_			1			
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
13J1 13J2							53 51		29.9 22.3	21 15	4.7 4.0			2130 7240		40.8 209	20.4 13.5			1.01 0.98					3.3 5.4	12.9 18.9	116 123	662 604
13L							93		22.4	4	2.9			15900		4.4	12.4			0.97					1.9	8.9	188	113
13M							86		23.2	3	3.3			15400		8.8	13.3			1.01					2.0	10.8	145	120
13N							60		16.9	7	3.5			8960		15.1	14.9			1.05					2.3	14.2	134	274
130							36		14.7	20	5.8			6830		218	15.6			0.28					10.1	27.1	97	646
13P							95		25.8	13	4.8			9520		4.1	17.6			1.21					1.8	11.1	157	349
13Rt 13Rm							119 38		12.1 5.8	11 4	3.9 1.3			3650 2910		35.6 23.9	19.1 4.7			0.38 0.18					1.4 2.2	16.7 8.7	98 41	416 94.7

![](_page_62_Figure_0.jpeg)

							_	Lab A	nalys	is (in µ	ug/g)							
										Ę							To RI	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	Praseodymiu	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
14(ash)t 14(ash)b	37.6 37.1		2.18 6.50		3.4 8.0		19.2 17.5		16.5 22.9			13.6 16.0				24.2 66.7	133 209	149 264
14A	65.1		3.60		6.9		27.9		36.2			17.7				25.8	213	255
14B	21.1		2.31		4.3		8.7		17.1			18.0				19.9	108	319
14F	57.7		3.63		6.6		25.2		30.1			16.1				28.7	196	381
14K	28.9		1.70		3.6		12.7		16.0			10.9				13.5	102	389

												Lab	Anal	ysis (in	µg/g	)												
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
14(ash)t 14(ash)b	2.37 4.66	98.5 86.7	9550 5410						13.8 10.6	а 8				28800 11300		23.8 120	7.3 7.0								2.0 27.4	8.8 34.1	75 63	128 254
14A	5.77	56.6	1580						23.1	6				15200		46.4	16.1								5.9	18.1	212	355
14B	3.00	295	811						4.6	2				17000		54.2	16.1								11.9	15.3	40	388
14F	9.53	58.0	600						17.1	12				6770		51.6	11.8								3.9	13.8	139	281
14К	3.19	41.7	706						5.6	4				2400		16.3	8.1								2.2	7.8	37	202

![](_page_64_Figure_0.jpeg)

								Lab A	nalys	is (in <sub>l</sub>	ug/g)							
										m							To Ri	tal EE
Sample ID	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymiun	Praseodym	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Whole coal	Ash
15A	55.0		2.54		6.0		26.0		32.7			13.5				19.7	180	204
15Bt 15Bm 15Bb	53.0 84.8 99.2	8.1 11.9 12.2	4.89 7.59 6.31	2.10 2.02 3.32	8.2 11.0 14.0	1.66 2.57 2.28	22.1 40.1 36.1	0.70 0.95 0.78	32.8 35.8 59.1	7.4 9.4 13.9	8.2 7.3 13.9	31.3 12.0 18.2	1.34 1.87 2.19	0.69 1.03 0.85	4.58 6.02 5.13	46.2 93.8 51.5	232 328 339	892 1432 575
15D	71.8		2.62		5.2		37.4		32.2			14.0				20.7	209	223
15E 15Ft 15Fb	71.9 40.3 46.1		3.05 2.21 3.26		5.6 3.8 5.7		34.3 20.3 20.5		33.6 18.9 24.7			13.9 14.4 12.8				23.1 18.6 25.6	212 136 163	245 236 361
15J	64.4		5.49		8.9		30.4		33.7			22.1				45.2	248	573
15Kt 15Km 15Kb	13.0 9.1 36.7		0.67 0.40 2.37		1.5 0.8 5.2		6.3 4.6 15.4		7.2 4.1 22.7			5.0 0.9 15.3				5.3 4.4 17.7	45 28 136	92 180 446

												Lab	Analy	ysis (in	µg/g	)										1		
Sample ID	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
15A	3.57	72.7	1580						20.6	6				11400		11.0	17.0								1.5	5.8	114	271
15Bt 15Bm 15Bb	1.94 0.90 8.29	15.2 12.3 64.5	1590 320 3640						19.9 13.9 18.8	3 3 8				13900 12100 7580		18.9 16.3 82.8	19.3 17.9 12.5								2.5 5.3 5.9	21.4 10.7 15.6	98 17 140	542 157 145
15D	2.88	6.3	591						28.3	3				14800		3.3	14.1								2.2	5.9	114	99.7
15E 15Ft 15Fb	3.55 8.00 3.61	12.6 48.2 29.1	971 228 417						26 17.9 14.8	5 10 12				14900 6130 8010		7.8 29.8 18.7	13.6 11.9 13.1								1.7 2.6 5.2	6.0 5.8 10.3	127 110 70	157 163 298
15J	16.6	72.7	368						17.6	14				7700		75.0	16.9								2.7	15.3	141	313
15Kt 15Km 15Kb	4.95 0.30 10.9	315 16.5 104	4060 222 626						4.3 4.1 12.6	4 1 27				1990 2680 3750		22.4 15.8 95.7	6.3 0.8 29.6								1.9 4.2 7.1	2.2 3.1 7.9	48 5 145	103 10.1 508

#### Appendix B - Analytical Results

Concentrations are reported on a whole rock basis as  $\mu g/g$  or parts per million

Sample ID	NDGS Field ID	Ash (wt%)	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	Praseodymium	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
1(sil)	225SIL	98.46%	41.1	1	.28		2.1		22.4		16.6			6.2				10.7	1.44	2.8	367	0.5		5.29	36	1.9	10.2	2	2.8		33.3	1580	38	1.4	26.1	49	34	1.78		6.8		6930	3.7	3.1	54	98
1B1 1B2	225A 184A1	64.01% 79.53%	191 124	18.8 1	0.1	5.37 2	21.8	3.6	83.1 57.6	1.33	97.7	24.1	22	30.6	3.34	1.38	8.72	88.7 53.2	7 33		4240	31		6 33	76 84		23	7	6.3 5.4		37.6	8610		78.0 47 8	15.5		420	0.83				2740	13.3	65.5 70.6	165 156	358 288
1F	225L	46.41%	38.8	3	.03	0.10	6.3		15.2	0.10	25.3			18.2		0.00	0.07	24.6	1.00		.2.10	0.1		0.00	55		10	15	8.7		01.0	8010		67.9	15.0			0.33				21.10	9.9	27.1	57	739
11	225B	57.54%	116	21.1 1	3.0	4.68	21.9	4.5	58.9	1.73	70.3	15.7	17	32.7	3.46	1.74	10.9	112							157		25	42	10			6400		94.4	26.8			0.67					3.4	27.9	405	1150
1lr 1k	225BR	95.02%	52.9	2	.29		4.4		29.1		24.4			15.2				20.5	1.99	16	3680	1.9		6.62	88	13	21	7	2.5		32.8	9090	157	4.5	13.4	96	173	0.97		9.8		4310	2.2	4.1	138	89.0
2A	184L	75.24%	61.5	5	.80		8.6		31.8		32.6			22.8				43.0	22	392	11700	5.4	0.7	6.6	81	100	25	14	10	0.1	27	18600	1310	139	28.4	87	587	0.74	0.2	12	2.0	2420	56.6	18.9	184	480
2B1b	184A3b	80.40%	205	19 1	0.4	5 2	21.6	3.7	97.6	1.31	98.9	24.4	21	39.9	3.4	1.4	8.8	80.0	6.9	90	1390	5.6	0.7	9.9	118	9.8	35	20	8.2	0.1	55	11300	35	75.5	19.5	108	353	1.05	0.20	28	2.3	3250	18.3	106	206	503
2B1t	184A3t	74.38%	203	18 1	0.2	5.9 2	23.1	3.5	91.9	1.44	112	26.4	25	31.0	3.4	1.5	9.4	76.1	13	329	9830	5.0	0.6	5.4	89	9.5	33	13	13	0.1	56	11400	56	233	22.0	41	523	1.23	0.4	22	2.4	2540	24.6	100	241	775
2B2 2Eb	184A2 184MB	69.90% 54.17%	361	28.5 1	38	8.67	35.4	5.4	148	1.97	1/5	12.4	38	41.5	5.22	2.06	13.4	132	9.39	285	11500	7.6	0.9	5.77	110	9.5	44	15	15	0.1	60.4	15000	22	159 42 2	29.1	48	548	1.54	0.6	31	2.5	2830	31.2	157	201	292
2Ft	184MA	41.38%	50.3	4	.82		8.9		19.1		32.5			15.2				37.9	9.59	266	1650					11			6.9			6030		78.5									15.2	23.1		541
2G	184N	66.75%	34.1	3	.86		4.0		17.5		15.7			12.0				33.8	6.22	43	1340					13			5.9			11800		35.7									2.8	26.9		299
21	184B	55.67%	301	28.4 1	5.5	9.42	34.0	5.5	107	1.91	172	40.1	41	30.9	5.00	2.11	13.6	130	25.0	270	581	2.9	0.3	4.38	102	17	17.5	29	6.2	0.1	20.7	5310	86	58.5	13.4	45	255	0.56	0.1	14.2	1.4	1960	5.7	42.4	160	533
2N 2N	184C 184D	44.98%	31.0 9.9	2.9 1	.73	0.75	2.1	0.6	5.2	0.26	15.0 5.7	4.0	3.4 1.6	0.9 11.9	0.48	0.25	2.04	14.0	4.88		775	1.9		0.14	19		6.3	10	3.3 2.4		9.7	4260		12.5	9.7		320	0.70				900 257	4.3	9.0	24	219
2Q	184E	47.92%	55.0	6.6 3	.94	1.82	7.5	1.3	21.0	0.58	33.3	7.7	7.8	12.9	1.14	0.56	3.79	31.1	10.5		978	1.9		3.36	44		9.8	14	3.0		7.7	3010		68.3	10.9		204	0.32				1140	9.1	19.4	93	392
2R	184F	32.69%	20.2	2.5 1	.36	0.79	3.0	0.5	7.9	0.18	13.5	3.1	3.3	5.9	0.44	0.19	1.20	11.8	6.11		986	1.1		0.87	25		4.3	14	1.9		3.4	1970		11.0	16.8		279	0.18				747	4.2	4.6	68	217
3A	208L	55.37%	53.3	40.0	9	^ 0.47	10.9	0.0	26.8	4.05	30.2	15.0		32.7	0.00	4.00	0.00	72.7	16.4	698	16800			5 50	00	57	00		19	0.4	10.0	23200	00	126	44.5	00	400	0.04		44.5		0450	83.5	44.4	007	870
3B10 3B1m	208AC	57.35% 69.34%	68.0	13.3	03	3.47	7.5	2.0	31.1	1.05	34.8	15.2	14	20.7	2.33	1.06	0.80	35.9	9.18	42.0	2780	3.8	0.6	5.52 6.37	50	6.3	32	6	4.8	0.1	46.0	5870	30	02.8 41.0	10.9	56	242	1.94	0.2	11.5	2.7	2380	8.4	32.7	126	193
3B1t	208AA	43.20%	141	14.7	8 4	4.27 ·	17.4	2.8	53.2	1.06	81.0	19.2	19	28.2	2.66	1.10	7.00	64.8	6.17	87.0	4860	4.4	0.3	1.37	72	11	17.3	6	7.1	0.1	15.4	10300	46	48.1	16.2	11	471	0.45	0.2	12.4	1.1	1270	10.8	60.7	107	334
3B2b	208A2B	78.17%	128	13.3	7.9	3.64	14.3	2.7	62.8	1.10	63.7	15.6	14	31.5	2.26	1.08	7.03	65.1	13.1	267	13800					9.8			7.6			8130		107									10.9	94.8		494
3B2t	208A2A	79.58%	122	10.8	5.2	3.15	12.4	2.1	55.3	0.85	60.4	15.1	13	26.4	1.91	0.86	5.52	51.4	8.66	186	6280	4.6	0.8	7.65	79	12	31	9	7.4	0.1	58.7	9100	43	85.8	15.9	64	268	1.47	0.3	19	2.5	2790	11.0	74.9	185	389
3B3	208A3	73 52%	73.3 90.6	91	5.5	2 68	9.9	19	33.2 42.0	0 79	30.3 44 3	10.9	9.6	19.2 21 7	1.55	0 78	5 12	40.0	6.03	86	7370	4.0	0.5	4.60	80	18	32	8	6.3	0.1	29.0 38.1	10300	113	87.8	14.7	57 69	299	1.09	0.2	13.6	2.0	3000	9.0 13.4	76.0	181	287
3Fb	208MB	41.01%	29.8	3	.75	2.00	4.7	1.0	14.5	0.70	14.9	10.0	0.0	17.4	1.00	0.70	0.12	32.7	6.18	33	1580	7.2	0.0	0.01		7.7	02		4.7	0.1	00.1	9170	110	44.1	10.4	00	200	1.10	0.1	10.0	2.0	0000	7.1	12.1	101	284
3Ft	208MA	36.71%	34.9	4	.70		6.5		15.9		20.6			14.1				41.7	7.06	49	1720					8.4			6.5			11300		75.2									12.7	24.5		418
3G	208N	68.67%	46.8	4	.34	10.0	5.3	5.4	23.9		21.6		47	15.8	5.04	4 70	40.4	38.5	8.10	35	3210	5.4		5.07	107	31	00		5.2		00.5	15900	00	37.9	0.7	50	100	0.57		00.4	10	0400	2.3	22.7	005	266
311	208B 208B3	55.97% 82.76%	314 161	30.0 1 15.5 7	3.1	10.9 5.33	38.6 19 1	5.1 2.7	95.1 63.6	1.44	183	41.7 21 7	47 23	40.7 26 1	5.84 2.98	1.70	10.4	101	19.3	66	376	5.4	0.4	5.97	167	23	30	23	3.3	0.1	29.5	5210	86	64.0 8.7	9.7	56	199	0.57	0.1	22.4	1.6	2190	2.3	81.4 22.6	285	489
3J	2080	90.29%	51.8	10.0	.91	0.00	3.4	2.1	26.9	0.00	21.8	- 1.7	20	18.0	2.00	0.00	0.00	15.6							90		26	2	3.4			11000		2.3	11.6			0.94					2.0	4.5	150	124
3K	208C	74.68%	20.3	0	.83		1.5		10.2		8.9			8.7				6.8							21		22	3	3.8			7010		1.5	3.8			1.26					0.7	6.2	37	96
4B1b	209AB	87.13%	137	15.1	9	4.00	16.5	3.1	66.6	1.09	71.9	17.4	16	29.0	2.58	1.17	7.13	81.2	11.9	142	11200	07	0.0	0.07	05	12	22	0	6.7	0.4	CO 4	10400	50	74.0	20.0	05	075	4.40	0.0	07	0.7	2020	16.7	73.8	240	271
4B1t 4B2	209AA 38B*	86.29% 70.13%	134	12.7	7.9	3.43	14.5 15.4	2.5	56.1 60.2	0.96	69.2 64 4	17.4	15 15	28.9	2.22	1.08	6.23 6.87	65.0	13.7	148	/4/	6.7	0.9	8.67	95	11	32	8	6.5	0.1	62.4	9990	58	125	20.9	65	275	1.46	0.6	27	2.7	3620	20.4	69.3	248	278
4F1b	209MB	44.24%	45.8	4	.17	0.00	6.9	2.0	19.6		25.0	.0.0		13.9			0.01	29.6	8.47	93	553					8.5			6.0			8840		108									16.4	34.7		401
4F1t	209MA	63.81%	46.0	3	.56		6.1		22.3		25.5			14.5				29.4	8.13	152	1860					10			7.5			9780		62.3									14.9	26.6		506
4F2	38A*	41.59%	44.7	7.8 4	.65	1.64	7.6	1.6	17.2	0.67	26.4	6.2	6.6	13.4	1.28	0.66	4.27	33.0	7.04	50	4180	3.4		1.47	15	21	12.2	6	8.4		5.4	9420		99.8	3.6		434	0.17				506	19.3	21.6	58	617
4G1 4G2	209N 38C	59 21%	34.3	4	.82		4.9		15.8		14.9			14.4				38.3	5.35	52 31	9280 6980					98			7.3			17800		32.0									3.0	23.0		395
411	209B	76.37%	224	20.6	9.5	6.41	24.3	3.6	96.9	1.00	112	28.4	26	37.0	3.85	1.23	7.22	88.5	7.65	36	5560	6.2	0.6	6.75	118	7.1	24	5	2.9	0.2	39.6	5380	69	14.2	9.5	68	190	0.65	0.10	29.6	1.8	2810	2.2	44.5	198	137
412	38D	81.95%	216	20.0 8	3.9	6.29 2	24.3	3.5	97.0	0.94	111	27.4	26	32.4	3.82	1.15	6.75	84.8	8.13	18	509	5.4	0.7	7.53	136	8.4	27	5	3.2	0.2	49.1	5190	63	13.4	10.8	81	85	0.77	0.1	25.7	2.0	3390	2.2	34.4	205	168
4J	209O	91.43%	52.6	2	.04	0.15	4.0	17	25.7	0.70	23.6	70	0.2	17.0	1 45	0.60	4 45	16.5	2.18	12	937				_	6.1	_		3.3		_	9790	_	4.4	_	_	_		_				1.7	5.9	_	123
4KD 4Kt	209CB 209CA	60.45%	32.0	0.0	+.9 .87	2.15	0.9 3.5	1.7	14.4	0.70	17.3	7.9	0.3	12.2	1.45	0.09	4.45	16.2	4.16	37	845					3.7			5.2 5.0			6540		7.0									4.0	5.2 7.4		241
5(nod)	224NOD	68.41%	11.8	0	.62		1.2		8.1		5.7			2.9				7.2		0.	0.0				20	0.1	4.6	<1	0.4			3930		<0.2	2.7			0.25					0.1	1.0	24	15
5Bb	224AB	84.35%	121	10.9	5.2	2.99 ′	12.3	2.2	61.5	0.76	59.1	14.9	12	21.2	1.95	0.84	5.08	57.0							71		31	8	4.8			11700		86.7	13.3			1.42					11.6	111	189	228
5Bt	224AA	80.90%	251	20.3 1	0.5	6.01 2	25.2	3.7	112	1.27	130	32.5	28	32.9	3.79	1.39	8.47	89.0							86		29	9	7.5			14000		142	18.2			1.24					18.5	99.8	194	297
DE 5E1	224Q 224M	78.11% 41.80%	42.6	13.5 0	90	4.23	9.6	2.5	40.5	0.87	40.1	17.1	20	23.2 14 3	2.59	0.89	5.87	49.3							36		83	4	6.0			6150		29.8	9.7			0.28					3.0 21.1	17.4	63	345
5F2	37L6	41.37%	46.6	9.8 5	5.1	2.81	12.3	1.8	14.1	0.66	48.7	9.1	13	19.8	1.79	0.68	4.50	36.8							00		0.0	-	0.0			0100		00.1	0.1			0.20					21.1		00	010
5G	224N	43.25%	125	14.6 8	3.6	2.86	14.5	2.9	52.5	1.17	57.8	14.4	13	19.3	2.44	1.17	7.64	74.1							53		17.3	4	5.4			5110		50.1	16.6			0.74					2.6	15.5	84	261
5lb	224BB	88.42%	142	10.0 4	.87	3.10	12.2	1.8	68.6	0.58	65.5	16.8	14	21.3	1.87	0.66	4.03	45.7							100		25	4	3.1			7510		7.2	13.2			0.90					2.3	10.6	167	124
oim 5lr	224B1 224BR	95.38% 92.40%	79.8	2	.38 88		0.U 3.6		38.8		35.3			12.0				21.9							00 116		17.8	3	2.3			4500		4.3	11.5 14 2			0.83					1.5	4.4 3.8	207	79 98
5lt	224BA	85.00%	141	11.3	5.5	3.36 <sup>·</sup>	13.6	2	67.6	0.65	65.1	16.6	14	23.6	2.09	0.73	4.38	52.7							117		35	5	3.0			6430		13.0	12.6			0.83					2.7	11.7	209	136
5K	224C	48.66%	29.9	1	.77		3.7		13.7		16.5			7.5				16.8							24		12.8	8	4.4			5880		7.8	12.2			0.91					2.7	5.6	50	281
5L	224P	75.45%	37.5	2	.39	0.74	3.3		19.6	1.05	17.2		45	16.2	0.00	4.07	0.50	20.5							98		24	28	4.8			18200		45.9	23.9			0.74					2.0	19.6	219	477
6B2	37L5A 38B2	88.23%	143	13.2	1.1	3.74	14.9 12.0	2.6	65.9 47 Q	1.05	69.1 53.3	17.0	15 12	30.0 24.4	2.29	1.07	6.59 5.61	67.0 54.8	10.7	94	5740	45	0.9	8 32	122	26	37	13	6.3 6.4	0.1	58.3	10600	112	105	20.1	96	370	1.18	02	21	2.6	3000	20.3	63.9	268	302
6B3	223A	73.27%	159	17.4 1	0.5	4.36	18.6	3.6	78.1	1.39	77.6	19.0	17	30.9	2.97	1.43	8.98	105	10.1	04	0140	4.5	0.0	5.02	70	20	32	7	7.8	5.1	0.0	13000	112	178	17.4	00	510	1.26	0.2	21	2.0	5005	36.1	131	196	442

Sample ID	NDGS Field ID	Ash (wt%)	Cerium	Dysprosium	Erbium	Europium	Gadolinium		Holmium	Lanthanum	Lutetium	Neodymium	Praseodymium	Samarium	Scandium	Terbium	Thulium	Vtt orbitum	TILEIDIUII	Yttrium	Antimony	Arsenic	Barium	Beryllium		Bismuth	Cesium	Chromium	Cobalt	Gallium	Germanium	Hafnium	Indium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium	Tantalum	Tellurium	Thorium	Tin	Titanium	Tungsten	Uranium	Vanadium	Zirconium
6E 6F1 6F2 6F3 6G 6I	214M 37L4 214M2 37L5 214N 214B	65.60% 32.04% 44.71% 50.07% 46.22% 90.75%	165 19.6 47.2 83.7 114 84.0	20.0 4.5 10.8 12.7 13.7	9.70 2.99 7.5 6.6 8.3 3.41	0 6.2 0 0.8 1.7 3.2 2.6	26 25 36 3. 72 9. 24 14 53 13 6.	.9 9 .6 .2	3.6 1 2.4 2.4 2.8	59.8 7.9 19.5 30.1 54.6 41.5	1.18 0.42 1.09 0.82 1.15	119 11.6 26.7 57.9 53.9 36.5	27.0 2.7 6.3 12.9 13.6	30 3.1 6.8 14 5 12	28.1 9.7 19.0 19.4 21.0 23.1	2 3.8 0.6 5 1.6 4 2.2 0 2.2 2	5 1.2 3 0.4 0 1.0 3 0.8 5 1.1	8 8. 2 2. 6 6. 7 5. 6 7.	16 7 66 2 76 7 47 5 47 7 2	25.9 28.9 21.0 57.8 23.6 28.1	15.4 7.78 9.29 4.50	68 113 55 34	547 3050 600 1850	) 3. <sup>-</sup> 6.:	7 5	1	.64 .19	13 23 27 106	32 8.3 30 17	10.7 17.5 8.9 28	6 6 5 5	8.7 4.7 6.5 5.9 6.8 4.5		6.1 8.7	11900 8550 9600 7030 6670 12500		51.3 55.8 92.6 77.2 26.1 6.4	4.0 5.5 5.2 13.9		318 400	0.12 0.19 0.20 1.03				436	4.8 8.9 6 10. 7.3 2.8 2.2	5     49.       9     14.       0     34.       8     16.       5     23.       2     15.	2 9 59 7 80 5 47 7 6 202	524 297 426 385 320 2 254
6lr 6J 6K1 6K2b 6K2t 7(bar)	214BR 214O 214C 214C2b 214C2b 214C2t 223BAR	92.77% 85.11% 58.88% 23.52% 67.22% 98.49%	61.9 49.4 19.6 102 35.9 12.6	14	2.37 2.30 1.03 7.52 1.91 0.57	3.	5. 4. 1. 8 15 3. 1.	0 1 8 .0 7 2	2.6	32.1 24.0 9.5 39.1 17.9 6.6	1.03	28.5 22.6 9.5 61.7 18.5 6.3	14.1	15	15.0 18.0 7.6 25.8 11.0 3.1	) 5 3 2.4	1.1	1 6	2 1 8.7 5 1	1.2 8.4 8.8 52.9 4.1 5.3	6.80 4.30 0.29	24 38 1.7	3940 7510 3200	0 0.4	4	2	70	84	5.9 2 2.3	21 22 18 5.1	3 11 7 1	2.8 4.2 3.2 0.6		7.8	20500 13500 6730 4280 6630 3150	84	1.7 11.2 8.3 9.8 9.8 0.5	3.2	28	524	0.24		1.9		930	2.1 1.9 1.1	2.9 12. 5.0 8.0 10. 5 1.1	9 128 3 0 - 3 - 7 27	<ul> <li>91</li> <li>214</li> <li>158</li> <li>487</li> <li>204</li> <li>20</li> </ul>
7B 7D 7E1b 7E1t 7E2 7F1	214A 37O 223R 223Q 37N 37L3	82.78% 27.05% 66.11% 63.27% 52.85% 37.10%	151 40.2 91.9 183 216 95.5	14.6 7.4 11.4 25.0 30.8 17.6	8.4 4.74 5.7 12.0 13.9 8.8	4.1 3.8 6.9 9.8 4.7	11 16 41 7. 30 15 96 30 35 40 75 21	.9 .0 .4 .0	2.9 1.6 2 4.5 5.3 3.2	69.8 18.4 31.7 58.8 61.1 28.7	1.16 0.70 0.75 1.44 1.71 1.13	76.8 23.6 72.2 131 175 81.8	18.7 5.4 15.3 28.8 37.6 16.8	17 5.9 18 31 5 46 21	28.0 12.0 27.1 23.0 33.3 20.9	2.5 1.1 2.2 4.7 3.5.7 3.1	9       1.1         6       0.6         1       0.7         0       1.5         9       1.8         5       1.1	7 7. 67 4. 6 4. 8 9. 4 1' 7 7.	43 7 33 4 98 4 83 9 1.5 9 48 6	9.8 1.3 2.9 4.1 3.0 8.8	10.1 22.6 11.2	164 107	5630 4480 1130	) 6.0 ) 5.0 0 3.0	6 0 6 1 6	).9 7  .2 4 (	.08	99 107 89 84 30	15 45	33 18.0 34 15.3 8.1	8 6 16 14 10	7.4 7.3 10 11 6.6	0.1	76.6 23.2 5.9	8300 10800 12200 9430 7790	33	96.7 51.9 124 131 110	20.9 14.5 31 25.8 6.4	82 46	322 163 203	1.54 0.70 0.80 0.56 0.16	0.30	16.0	2.6	299 180 520	0 20. 3.8 12. 0 6.4 0 17.	7 58. 3 26. 6 34. 1 25. 1 25.	8 214 5 234 9 382 2 38 0 74	1 359 3 404 2 720 7 839 4 372
7F2b 7F2t 7F3 7F4b 7F4t 7F5	223MB 223MA 37L2 37M* 37L* 37L7	40.53% 44.90% 40.07% 54.63% 40.07% 41.22%	69.9 18.0 84.8 72.2 114 44.6	12.4 10.3 17.3 7.5	6.5 1.99 6.4 5.9 8.7 4.20	3.5 2.2 4.6 1.8	10 3. 55 15 24 10 67 20 38 8.	.2 2 .4 .0 .0 4	2.3 2.1 3.2 1.5	29.6 6.7 23.7 28.6 36.2 17.0	0.81 0.81 1.13 0.61	37.0 12.7 62.6 39.3 80.1 33.1	13.5 9.6 17.7 7.0	i 16 9.3 20 8.3	23.0 7.4 19.3 18.9	3 2.2 3 2.2 3 1.7 2.9 7 1.2	5 0.8 1 0.8 3 1.2 7 0.5	4 5. 5 5. 2 7. 9 3.	49 4 29 5 43 6 82 3	51.7 7.9 5.3 55.0 4.0 3.5								38		13.8	4 6	4.2			9040		33.5	13.7 6.5			0.88					3.8	3 19.	7 92 3 40	199 324
7G1 7G2 7HTm1 7HTm2 7HTm3 7J1	223NII 223N 184G 184H 184I 37K*	63.27% 68.14% 49.74% 30.10% 17.60% n/a	38.9 21.5 9.3 7.1 45.8	3.3	5.2 3.67 1.64 0.47 0.36 2.20	, , , , , , , , , , , , , , , , , , ,	6. 4. 2. 0. 0. 34 3.	7 4 8 8 7 5	0.7	22.0 12.8 6.4 4.0 23.5	0.37	23.6 16.0 11.5 3.9 3.5 21.3	5.3	3.8	10.3 10.3 5.0 1.8 2.4	0.5	1 0.3	5 2.	4 3 1 ( 35 35	8.9 4.4 8.2 6.0 3.9 7.0	1.94 0.55 0.46	37 11 5.1	2030 1620 1780	) 1.0	6 0	).2 1	.71	49 47 23	6.1 2 3.1	12.7	7 4	4.7 4.5 1.3 0.8 0.7	0	11.9	9030 6300 4840 4580 3800	118	28.5 31.4 6.0 4.8 2.3	20.7 17.9 5.5	28	330	0.82	<0.1	1 3.5	0.8	123	3.1 3.1 0 1.6 1.1	6.4 6 4.6 1 1.1	7 150 4 140 5 35 7 1	) 212 ) 193 64 28 25
752 7Kb 7Kt 7M 7N 70 70	223B 37J* 37F* 223V 37E* 223W	n/a 59.28% 68.96% 29.37% 92.81%	35.3 26.7 57.9 19.3 99.6	4.0 3.4 5.2 10.1	2.35 2.26 1.92 3.34 3.95 5.5	0.8 0.8 0.8 2.8	4. 14 4. 36 3. 5. 38 3. 39 11	3 6 4 7 .9	0.8 0.7 1.2 1.9	33.8 15.3 12.7 28.1 9.6 44.1	0.31 0.28 0.64 0.75	28.0 18.8 15.9 26.9 10.4 54.6	4.5 3.9 2.4 12.9	4.4 3.8 2.7 12	19.	0.6 0.5 0.6 0.6 5 1.8	5 0.3 4 0.2 3 0.6 3 0.7	3 2. 9 1. 1 3. 6 4.	02 1 85 1 93 3 84 4	7.0 5.0 5.1 4.0								88 56 70		20 16.0	46	4.0 3.1			7390 6580		1.7 19.6 8.5	20.4 13.8			0.90					1.5	7 15. 6 5.	0 200	) 360 3 237
7P 7Q 7R 7S 8(crb)	223WR 223X 223Y 37A* 223Z 215Y	95.20% 52.12% 58.77% n/a 71.41% 95.40%	48.6 31.7 28.5 54.8 66.2	3.0	2.80 3.19 3.26 1.47 2.77 2.78	1.1	5. 6. 3. 13 3. 5. 5.	7 2 8 7 3	0.6	19.3 17.6 11.4 23.4 36.0	0.18	29.7 14.5 18.2 28.8 29.1	4.1	4.4	13.4 14. 16.4 16.9 12.7	0.5	3 0.2	20 1.	2 2 25 2 2 2 2 2 2 2	25.3 26.2 2.0 2.0 2.0	3.15	19.0	618					49 65 81	37	20.0 11.1 19.9 19.4	25 59 36	3.5 3.5 4.1 2.8 3.0			4910 6950 7860 8000		2.3 66.7 24.8 18.8 2.9	18.1 30 21.4			0.39 0.53 0.59					1.0 11. 3.4 1.0	4 17. 4 14. 5 16. 9 8.4	6 97 7 17 6 21 4	1 135 634 1 469 1 237 116
8(1e) 8A 8B 8C 8D 8F1	215Fe 215L 215A 215X 215Z 37L8	89.67% 71.56% 75.42% 85.01% 91.54% 32.97%	20.9 120 101 163 74.4 42.5	19.5 9.1 21.0 8.8	1.34 13.1 5.4 12.5 4.03 4.89	4.2 2.7 4.7 2.2	25 19 73 11 76 23 6. 24 10	4 .2 .1 .4 5 .0	4.3 1.8 4.5 1.7	10.9 61.6 46.6 79.7 39.6 14.5	2.00 0.78 1.50 0.70	10.3 66.3 51.7 84.1 34.1 36.7	15.2 12.5 20.0	9.9	37.5 20.1 32.4 17.5 17.5	5 3.14 2 1.6 4 3.6 3 1 1.5	4 1.8 5 0.7 0 1.6 2 0.6	6 12 4 4. 0 9.	2.3 1 77 4 38 1 38 3 59 3	4.1 128 9.5 143 15.3 4.0	15.9 5.59 14.0 7.38 6.74	152 464 146 33 34	6020 4940 724 6920 3580	) ) 3.0 ) 1.9	o c 9	0.5 5	.02	74 29	253 7.1 15 20	21 6.4	8	1.2 12 7.4 7.3 6.6 5.6	0.1	39.4 2.6	2030 23400 12500 13000 17400 10400	230	31.3 39.3 48.8 8.5 11.7 73.1	22.2 5.3	60	360 132	1.06	0.20	) 14.4	1.9	282	2.3 36. 20. 3.1 2.6 14.	6 17. 0 44. 1 27. 6 12. 6 19.	9 6 3 15 2 1 9 55	50 599 337 545 371 545 345
8G 8I1 8I1r 8I2 8I2r	215M 215B 215BR 215BR 215B2 215B2R 215O	40.33% 66.69% 83.31% 91.92% 58.94% 92.65%	30.3 33.7 189 46.5 352 146 56 8	16.5 34.8 7.1	4.64 7.4 1.65 16.7 3.58	5.5 11 2.8	52 20 3. .5 42 30 9.	6 .7 .2 9	2.8 6.2 1.3	10.4 21.6 78.0 24.3 129 69.8	0.79 1.91 0.43	14.6 101 20.3 199 67.1	24.4 46.7 17.4	24 49 13	10.0 11.1 30.0 19.4 35.0 19.1	2 3 3.1 4 0 6.4 2 1.4 0	3 0.9 6 2.2 2 0.4	95 5. 11 13 7 3.	82 6 1 3.4 1 03 3	7.5 5.3 3.8 143 1.1	4.31 7.95	24 23	6200 541	3.0	6 0	0.6 8	.42	117 104 122 103	12 13 7.9	25 26 34 28	6 2 20 4	4.7 4.7 3.3 2.6 4.5 2.5	0.1	40.4	7870 6450 7100 5240 6390	56	18.9 14.1 8.9 47.0 2.9	11.6 12.2 9.7 12.8	87	107	0.88 0.78 0.53 0.83	0.1	45.9	2.2	355	0 2.5 2.7 2.2 2.2	8.8 5 39. 5 5.1 2 60. 2 4.0	4 209 7 18 4 230 5 17	332 208 3 167 1 80.0 0 391 2 77 2 181
8K 9B 9F 9I 9Ir 9K	215C 226A 226L 226B 226BR 226C	53.24% 65.69% 44.54% 38.57% 92.00% 63.01%	37.7 237 44.4 268 226 26 4	24.3 31.7 14.8	2.52 13.3 4.49 15.7 7.5	9.5 5.3	4. 55 27 7. 52 39 32 19 2	7 .5 5 .8 .1	4.7 5.9 2.7	17.0 100 17.5 110 101	1.74 1.95 0.90	20.4 122 27.2 160 111	29.9 34.4 28.2	28 37 23	9.9 31.3 15.9 34. 20.4	3 4.20 9 1 5.81 1 2.84	3 1.7 2 2.0 4 1.0	'9 11 17 12 10 6.	2 1.4 1 2.8 1 2.8 6 1	9.3 1.8 111 9.1 125 8.8 1.6								34 85 35 98 117 25		11.6 19.8 16.0 22 33 14.3	11 10 9 15 8 7	2.9 8.1 9.7 6.7 2.9 2.9			5270 15300 4590 2360 5960 6090		8.7 175 69.1 134 9.2 3.1	6.4 20.2 10.4 18.2 13.3 4 1			0.50 0.50 0.23 0.26 0.94 0.67					1.2 22. 14. 4.7 2.4	2 4.0 2 20 9 26. 7 51. 10.	5 72 0 15 3 99 3 21 4 18 5 62	194 5 413 741 6 697 9 110
10(crb) 10A 10B 10F 10G 10J	228L 228X 228A 228Y 228N 228N 228O	79.96% 89.02% 75.31% 61.54% 88.67% 91.37%	54.0 103 101 51.3 88.5 49.1	11.2 9.3	2.69 5.7 5.2 3.03 4.33 1.91	3.2 2.5	4. 29 13 59 10 5. 7. 3.	9 .6 .2 8 3 2	2	23.2 35.0 43.6 23.2 43.6 24.8	0.74 0.72	24.4 66.6 51.1 27.6 39.7 20.9	15.3 12.7	15	17.3 15.9 22.0 13.3 13.9 17.1	5 9 2.04 6 1.62 8	3 0.7 2 0.7	78 5. 72 4.	12 4 61 4 3 3 1	21.8 0.7 5.8 6.4 7.0 5.2								59 80 113 55 48 85		21 27 25 16.4 34 27	12 13 4 12 4 3	9.8 5.8 4.6 5.6 5.7 3.6			20500 17400 7750 7760 5860 10700		71.6 13.6 27.9 73.9 21.2 6.3	23.1 21.3 14.3 8.0 29.5 12.5			0.88 0.96 1.05 0.54 2.83 1.08					30. 5.9 3.9 6.2 7.7 2.2	1 25. 22. 46. 2 28. 7 12. 2 7.0	0 103 0 230 6 249 4 133 1 95 0 149	3 539 5 280 9 222 3 422 5 235 5 148

Sample ID	NDGS Field ID	Ash (wt%)	Cerium	Dysprosium	Erbium	Europium	Gadolinium	Holmium	Lanthanum	Lutetium	Neodymium	raseodymium	Samarium	Scandium	Terbium	Thulium	Ytterbium	Yttrium	Antimony	Arsenic	Barium	Beryllium	Bismuth	Cesium	Chromium	Cobalt	Germanium	Cellianun	Hafnium	Lithium	Magnesium	Manganese	Molybdenum	Niobium	Rubidium	Strontium Tantalum	Tellurium	Thorium	Tin	Titonium	Tungsten	Uranium	Vanadium	Zirconium
10K	228C	67.86%	32.1	-	1.21		2.6		16.5		15.4	<u> </u>		11.3	1	-		0.2	1	1		-			49	18	3.8 1	4	3.7		5940		7.7	13.0		1.2	2	-			2.	) 6.6	5 12	0 229
11B1	229A	88.12%	49.6		2.32		3.5		25.1		20.1			18.7			1	4.4							108	2	5 3	3	5.3		1160	0	33.6	14.0		1.2	7				3.0	6 14.	1 23	3 258
11B2	229N	85.83%	47.0		1.94		3.7		24.7		21.5			13.5			1	6.6							73	2	6 2	2	3.0		7600	1	6.3	9.3		1.2	2				1.9	7.7	14	7 132
11C	229Q	94.27%	74.6		2.81		6.2		35.0		34.6			10.8			2	25.3							50	17	7.8 2	2	2.7		8020	)	1.1	14.1		1.2	D				2.0	6 4.5	5 74	100
11J	2290	88.69%	39.9		1.76		2.8		20.2		17.4			11.5				4.3							71	16	6.4	7	3.0		1020	D	1.5	11.2		0.8	7				1.3	5 5.5	5 11	8 210
11K	229C	68.34%	19.9		1.04		2.2		9.1		11.2			9.5				8.6							25	15	5.8 1	1	3.0		6090	)	9.7	5.4		0.8	Э				2.0	6.3	3 48	186
12E	235F	65.21%	31.2		1.89		2.4		16.8		13.3			10.5			'	6.2							67	14	1.7	5	3.2		1130	0	11.3	13.5		0.7	)	_			2.	2 49.	3 22	8 287
12F	235E	89.86%	43.4		1.37		2.6		22.0		18.9			9.3			1	1.1							56	17	7.2	3	2.9		8350		2.4	13.1		0.9	3	4			2.	5 8.0	10	9 117
12G	235D	71.41%	28.8		1.14		2.1		12.6		12.2			1.5				1.0							13	14	1.0 4	4	2.8		1160	0	11.7	3.1		1.0	3	-			1.	29.	0 54	86
12L 12D	2350	70.71%	40.4		2.44		4.2		19.4		21.Z			15.4				7.9							73	4	2 0	20	5.0		0230		105	18.4		0.0	1	-			8.	0 10.	3 19	5 593 1 297
12P	2350	31 / 0%	20.6		4.90		9.5		7 2		42.3			10.4				2.0							20	4	5 6	6	2.3		2320		17.4	27 /		0.7	*				5.0	3 36	2 20	1 307
13.11	2344	88 38%	26.4		2.02		24		13.9		11 1			13.2				7.6							53			21	4.7		2130		40.8	20.4		1.0	1	-			3	3 12	9 11	6 662
13.12	23442	84 36%	39.1		2.84		3.8		21.1		17.5			16.7				7.8							51	2	2 1	5	4.0		7240		209	13.5		0.9	2				5.	1 18	9 12	3 604
13L	234B	93.12%	53.6		2.32		4.5		28.8		24.6			14.0				1.2							93	2	2 4	4	2.9		1590	0	4.4	12.4		0.9	7				1.9	8.9	18	8 113
13M	234C	94.09%	60.0		2.00		4.3		31.4		26.5			14.9			-	8.3							86	2	3 3	3	3.3		1540	0	8.8	13.3		1.0	1				2.0	) 10.	8 14	5 120
13N	234D	86.97%	41.4		1.83		2.8		22.7		18.1			10.3			1	5.3							60	16	6.9	7	3.5		8960	)	15.1	14.9		1.0	5				2.3	3 14.	2 13	4 274
130	234E	54.19%	27.4		2.37		3.5		14.2		14.6			11.5			2	1.1							36	14	1.7 2	20	5.8		6830	)	218	15.6		0.2	3				10	1 27.	1 97	646
13P	234F	86.91%	78.3		3.34		6.5		41.5		35.9			17.9			2	9.5							95	2	6 1	3	4.8		9520	1	4.1	17.6		1.2	1				1.8	3 11.	1 15	7 349
13Rm	234H	39.10%	17.1		1.09		2.2		7.4		10.5			5.3				9.3							38	5	.8 4	4	1.3		2910	)	23.9	4.7		0.1	3				2.3	2 8.7	41	95
13Rt	234G	50.08%	39.4		2.03		4.9		17.0		25.4			13.8			-	5.9							119	12	2.1 1	1	3.9		3650	)	35.6	19.1		0.3	3				1.4	1 16.	7 98	416
14(ash)b	237AshB	79.30%	37.1		6.50		8.0		17.5		22.9			16.0			6	6.7	4.66	87	5410					10	0.6 8	8			1130	D	120	7.0							27.	4 34.	1 63	5 254
14(ash)t	237Ash	89.17%	37.6		2.18		3.4		19.2		16.5			13.6			2	24.2	2.37	99	9550					13	3.8 3	3			2880	0	23.8	7.3							2.	8.8	3 75	, 128
14A	237A	83.46%	65.1		3.60		6.9		27.9		36.2			17.7			2	5.8	5.77	57	1580					2	3 6	6			1520	0	46.4	16.1				4			5.9	9 18.	1 21	2 355
14B	237B	33.87%	21.1		2.31		4.3		8.7		17.1			18.0			1	9.9	3.00	295	811					4	.6 2	2		_	1700	D	54.2	16.1				_			11.	9 15.	3 40	/ 388
14	237E	51.52%	57.7		3.63		6.6		25.2		30.1			16.1			4	8.7 9	9.53	58.0	600					1/	.1 1	2			6770		51.6	11.8				4			3.	9 13.	8 13	9 281
14K	23/G	20.21%	28.9		1.70		3.0		12.7		10.0			12.5				3.5	3.19	42	1590					0	.0 4	4			2400		10.3	0.1				-			2		5 3/	202
15Rb	236BC	58.06%	00.2	12.2	6.3	3 32	0.0	23	20.0	0.78	50 1	13.0	14	13.0	2 10	0.85	5 13 4	9.7	8.20	65	3640					15		8			7580		82.8	12.5				-			1.5	0 0.0	6 14	+ 271
15Bm	236BB	22 01%	84.8	11.0	7.6	2.02	14.0	2.5	10 1	0.70	35.8	0.4	73	12.0	1 87	1.03	6.02 0	3.8 (	n on	12	320					13		3			1210		16.3	12.0							5.	3 10	7 17	7 157
15Bt	236BA	26.06%	53.0	8.1	4 89	2.02	82	17	22.1	0.33	32.8	74	8.2	31.3	1.34	0.69	4 58 4	6.2	1.94	15	1590					19	9.9	3			1390	0	18.9	19.3							2	5 21	4 98	3 542
15D	236C	93.66%	71.8	0.1	2.62	2.10	5.2		37.4	0.10	32.2		0.2	14.0		0.00		0.7	2.88	6.3	591					2	8 3	3			1480	0	3.3	14.1							2.3	2 5.9	11	4 100
15E	236D	86.58%	71.9		3.05		5.6		34.3		33.6			13.9			2	3.1	3.55	13	971					2	6 5	5			1490	0	7.8	13.6							1.	6.0	) 12	7 157
15Fb	236EB	45.15%	46.1		3.26		5.7		20.5		24.7			12.8			2	5.6	3.61	29	417					14	1.8 1	2			8010	)	18.7	13.1							5.3	2 10.	3 70	298
15Ft	236EA	57.42%	40.3		2.21		3.8		20.3		18.9			14.4			-	8.6 8	8.00	48	228					17	7.9 1	0			6130	1	29.8	11.9				-			2.	5.8	3 11	0 163
15J	236F	43.24%	64.4		5.5		8.9		30.4		33.7			22.1			4	5.2	16.6	73	368					17	7.6 1	4			7700	)	75.0	16.9							2.	7 15.	3 14	1 313
15Kb	236GC	30.47%	36.7		2.37		5.2		15.4		22.7			15.3			1	7.7 '	10.9	104	626					12	2.6 2	27			3750	)	95.7	29.6							7.	1 7.9	14	5 508
15Km	236GB	15.47%	9.1		0.40		0.8		4.6		4.1			0.9				4.4 (	0.30	17	222					4	.1 '	1			2680	)	15.8	0.8							4.:	2 3.1	5	10
15Kt	236GA	49.09%	13.0		0.67		1.5		6.3		7.2			5.0	ļ			5.3 4	4.95	315	4060					4	.3 4	4			1990		22.4	6.3							1.9	9   2.2	2 48	103

\* REE data for 9 samples previously reported in Kruger and others (2017), pages 71-72.